A first-in-class anticancer dual HDAC2/FAK inhibitors bearing hydroxamates/benzamides capped by pyridinyl-1,2,4-triazoles

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ABSTRACT

Novel 5-pyridinyl-1,2,4-triazoles were designed as dual inhibitors of histone deacetylase 2 (HDAC2) and focal adhesion kinase (FAK). Compounds **5d**, **6a**, **7c**, and **11c** were determined as potential inhibitors of both HDAC2 (IC$_{50}$ = 0.09−1.40 μM) and FAK (IC$_{50}$ = 12.59−36.11 nM); **6a** revealed the highest activity with IC$_{50}$ values of 0.09 μM and 12.59 nM for HDAC2 and FAK, respectively. Compound **6a** was superior to reference drugs vorinostat and valproic acid in its ability to inhibit growth/proliferation of A-498 and Caki-1 renal cancer cells. Further investigation proved that **6a** strongly arrests the cell cycle at the G2/M phase and triggers apoptosis in both A-498 and Caki-1 cells. Moreover, the enhanced Akt activity that is observed upon chronic application of HDAC inhibitors was effectively suppressed by the dual HDAC2/FAK inhibitor. Finally, the high potency and selectivity of **6a** towards HDAC2 and FAK proteins were rationalized by molecular docking. Taken together, these findings highlight the potential of **6a** as a promising dual-acting HDAC2/FAK inhibitor that could benefit from further optimization.

1. Introduction

The approach "one agent multiple targets" or "multi-target agents" gained significant support in the field of drug discovery, and has come to be known as polypharmacology [1]. It is generally agreed upon that combination therapy has a considerable therapeutic relevance; however, such an approach is not without significant disadvantages. For example, patients taking combination therapy suffer from unpredictable pharmacokinetic profile, drug-drug interactions, and physical incompatibilities, leading to unpredictable and additive side effects and adverse reactions, unforeseen drug inactivation or precipitation of the formulation [2]. On the contrary, a single compound acting on dual or multiple targets can concurrently modulate various tumor pathways to get desirable therapeutic efficiency without such additive side effects [3]. Consequently, designing and developing multi-target drugs implemented an adequate strategy for cancer therapeutics.

Histone deacetylases (HDACs) are important therapeutic targets because their inhibition can reverse the aberrant epigenetic states linked to cancer. The several FDA approved inhibitors of HDAC for cancer treatment are vorinostat (SAHA), romidepsin, belinostat, panobinostat, and chidamide (Fig. 1) [4]. The HDAC binding pocket consists of a hydrophobic cap, linker, and zinc binding group (ZBG). Fortunately, each part of HDAC pharmacophore can be modified to

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allow for the hybridization approach. The cap group can be modified with a variety of hydrophobic moieties, the linker could be aliphatic or aromatic, and the ZBG could be a carboxylic acid, hydroxamic acid, 2-aminobenzamide, or thiol [5]. Among HDAC inhibitions, those that are selective for the HDAC2 isoform are particularly important because HDAC2 is overexpressed in several cancers such as prostate, gastric, non-small lung, colon, hepato-cellular, and renal [5−11]. In fact, selective inhibition of HDAC2 isoform inhibits tumor growth, promotes apoptosis, and results in G2/M cell cycle arrest and G1 delay [12,13].

Despite their demonstrated anti-tumor efficacy, most of the studied HDAC inhibitors are associated with the emergence of resistance upon prolonged administration, such as Akt kinase activation and STAT3 phosphorylation [14−19]. Several researchers embarked upon developing dual or multi-target HDAC inhibitors as a promising research area. Currently, there are several dual/multi acting HDAC inhibitors in clinical trials such as CUDC-907 (PI3K and HDAC), 4SC-202 (LSD1 and HDAC), tinostamustine (alkylating agent and HDAC), and CUDC-101 (EGFR, HER2, HDAC), each combining a diverse array of targets with HDACs [4,20,21].

Focal adhesion kinase (FAK) is an essential cell signaling regulator inside the microenvironment of the tumor. It plays an indispensable role in regulating tumor cell survival, migration, metastasis, proliferation, and angiogenesis. FAK is recognized as a promising therapeutic target as it is overexpressed in several tumors [22,23]. The human FAK gene is comprised of three domains: an amino-terminal FERM domain, central protein kinase, and a carboxy-terminal focal adhesion targeting (FAT) domain. Of these domains, the maximum effort has gone into targeting the kinase domain. The main pharmacophore of the FAK inhibitors is comprised of diphenyl five/six-membered nitrogenous heterocycles. The six-membered are the early discovered inhibitors and got the most attention, whereas the five-membered are growing as quite promising FAK kinase inhibitors. Compounds PF-573228 and TAE226 are potent FAK kinase inhibitors; however, they did not enter clinical trials because the former did not show remarkable antiproliferative activity, and the latter affected the glucose blood levels and glucose metabolism (Fig. 2) [24]. Four FAK inhibitors completed phase I clinical trials CEP-37440, PF-562271, PND-1186, and GSK-2256098, while defactinib (VS-6063) completed phase II clinical trials [25]. Meanwhile, the five-membered FAK kinase inhibitors revealed a substantial ability to inhibit FAK and tumor cells (Fig. 3) [26]. Several discovered five/six-membered FAK kinase inhibitors revealed dual/multi-kinase inhibitory activity due to the various kinases' structural similarities. However, till now, there are no reported dual acting FAK inhibitors that target another non-kinase protein.

1,2,4-triazole and pyridine rings constitute useful heterocyclic moieties in developing several anticancer agents [27,28]. Besides their known antiproliferative activity, 1,2,4-triazole and pyridine bearing compounds showed potent FAK and HDAC inhibitory activities. Moreover, pyridine moiety was widely used as a surface recognition group in developing potent HDAC and FAK inhibitors as well [5,26,29−33].

The administration of HDAC inhibitors is associated with Akt kinase activation, leading to pro-survival cell elicitation effects, whereas FAK kinase inhibition leads to Akt suppression [14,15,34]. All the reported dual/multi-acting FAK inhibitors targeted FAK and another different kinase; these multi-kinase targeting compounds are prone to have side effects. Because of this, the best is to merge non-kinase targets such as histone deacetylase with FAK kinase; unlike the non-selective multi-kinase inhibitors, derivatives that inhibit non-kinase proteins are expected to show high levels of specificity.

Given this underwhelming performance of HDAC and FAK inhibitors created to date, we set out to design and synthesis of 20 novel dual-acting HDAC2/FAK inhibitors that leverage the promising properties of 1,2,4-triazole and pyridine heterocyclic rings.
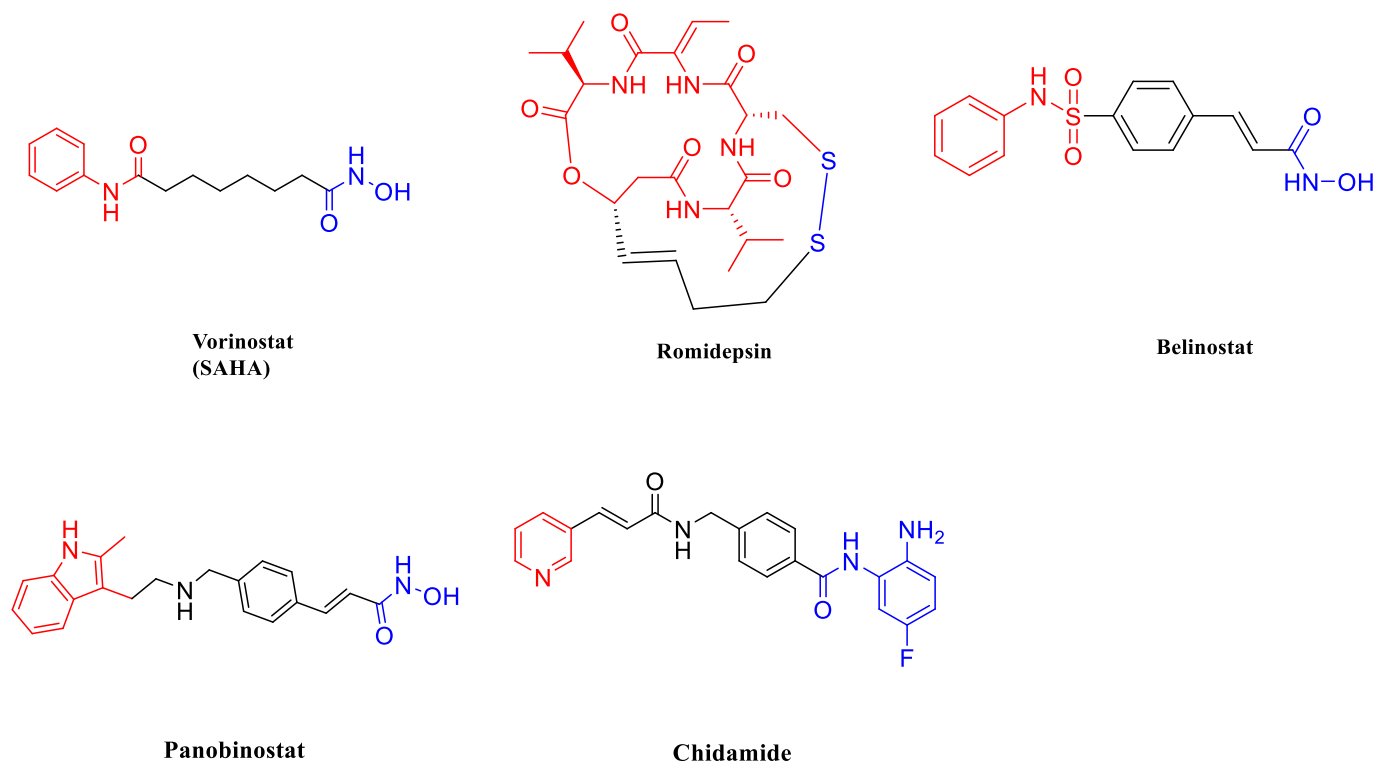

Fig. 1. Chemical structures of the FDA approved HDAC inhibitors.

The design of the dual inhibitors involved using 5-pyridinyl-1,2,4-triazole moiety as the surface recognition group and five carbon atom linker (aliphatic or aromatic) was implemented to the designed structures, while the ZBG was carboxylic acid, hydroxamic acid, or 2-aminobenzamide (Fig. 4). Moreover, the variation of substituents (methyl, ethyl, allyl, or phenyl) was made on N^4 of the 1,2,4-triazole ring Table 1.

2. Results and discussion

2.1. Chemistry

The synthesis of the 5-pyridinyl-1,2,4-triazole derivatives **5a-d**, **6a-d**, **7a-d**, **10a-d**, and **11a-d** (Table 1) is outlined in Scheme 1. Isonicotinic acid hydrazide **1** was treated with the appropriate isothiocyanate derivative to give thiosemicarbazide **2**. Cyclization of **2** was accomplished through the addition of 2 N NaOH followed by acidification with conc. HCl to afford the corresponding 1,2,4-triazole-3-thiol derivatives **3a-d** [35]. Alkylation of **3a-d** with ethyl-6-bromohexanoate or ethyl-4-bromomethyl benzoate in the presence of triethylamine afforded the ester intermediates **4a-d** and **8a-d**, respectively. Hydrolysis of the ester derivatives using LiOH afforded the corresponding carboxylic acid derivatives **5a-d** and **9a-d**, respectively. The acid intermediates **5a-d** and **9a-d** were treated with *N,N'*-carbonyldiimidazole (CDI) in acetonitrile at room temperature. The resultant imidazolide intermediate was either treated with hydroxylamine hydroxide to yield the target hydroxamic acid derivatives **6a-d** and **10a-d** or with 2-phenylenediamine in the presence of trifluoroacetic acid to afford the corresponding 2-aminobenzamide derivatives **7a-d** and **11a-d**, respectively [30,36]. Chemical structures of the synthesized 5-

pyridinyl-1,2,4-triazole derivatives were confirmed by ^{1}HNMR, ^{13}CNMR, and mass spectroscopy and their purity were checked elemental analysis.

2.2. Biological investigation

2.2.1. In vitro antiproliferative activity

The synthesized compounds were evaluated for their anticancer activity as a single dose (10 μM) against six different cell lines derived from three tumor subpanels, including colon (HCT116 and HT-29), leukemia (K562, and KG-1), and renal (A-498 and Caki-1) cell lines, in which both HDAC2 and FAK are overexpressed, using the MTT assay (Table 1) [37–39]. Most compounds showed promising anticancer activity with growth inhibition percentages. Among them, compounds **5d**, **6a**, **7c**, and **11c** were the most potent against all the tested cells (GI% = 40.12–99.81%). The renal cancer cell lines A-498 and Caki-1 were the most sensitive toward **5d**, **6a**, **7c**, and **11c** (GI% = 89.34–99.81%). Therefore, we tested the effect of different concentrations (1, 5, 10, 25, 50, 100, 1000, 2000, and 4000 μM) of the synthesized compounds on renal cancer cells (A-498 and Caki-1), using DMSO as a negative control, whereas VPA, SAHA, and TAE226 were used as positive controls. The half maximal inhibitory concentration (IC$_{50}$) was calculated for each compound (Table 2). Compounds **5d**, **6a**, **7c**, and **11c** displayed significant potency (IC$_{50}$ = 0.95–2.35 μM) against the tested renal cancer cell lines. Interestingly, **6a** showed a substantial ability to inhibit renal cancer growth; its IC$_{50}$ values were comparable to TAE226 and superior to VPA and SAHA. On the other hand, compounds **7a**, **5d**, **10a**, **10c**, **10d**, **11a**, and **11d** showed moderate potency (IC$_{50}$ = 6.19–20.85 μM), whereas the derivatives **5b**, **5c**, **10b**, **6b**, **6c**, **6d**, **7d**, and **11b** showed low potency (IC$_{50}$ = 76.36->100 μM). These

PF-573228   **TAE226**   **CEP-37440**   **PF-562271**

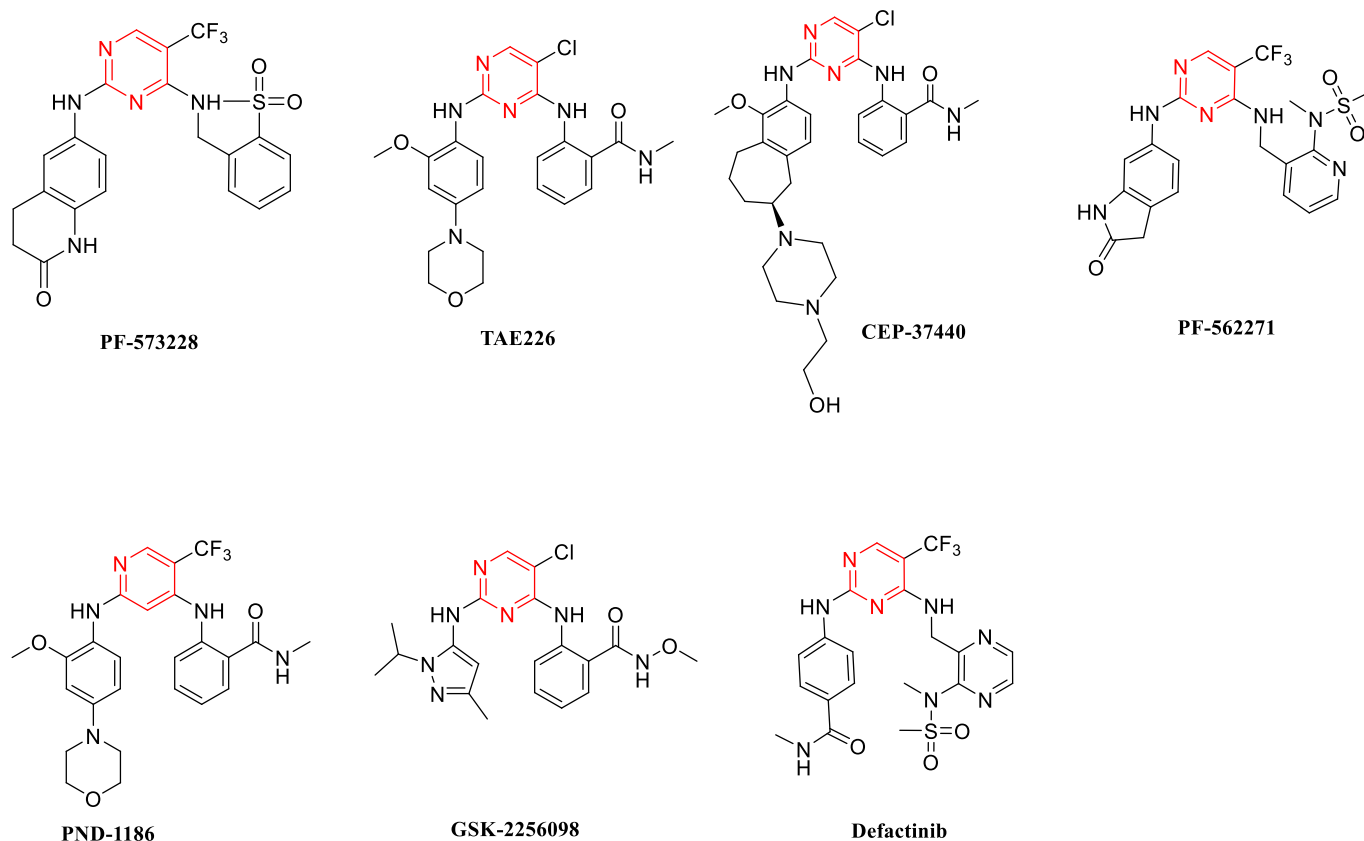**PND-1186**   **GSK-2256098**   **Defactinib**

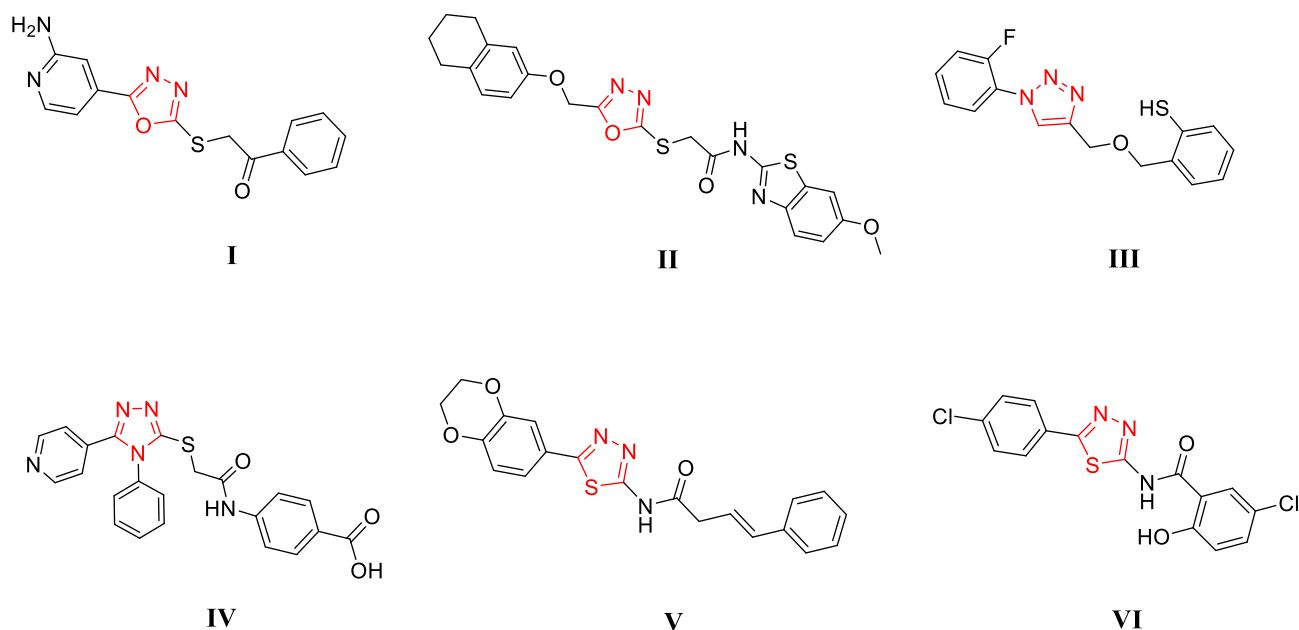**Fig. 2.** Chemical structure of the six-membered FAK kinase inhibitors.

Fig. 3. Chemical structure of the five-membered FAK kinase inhibitors.
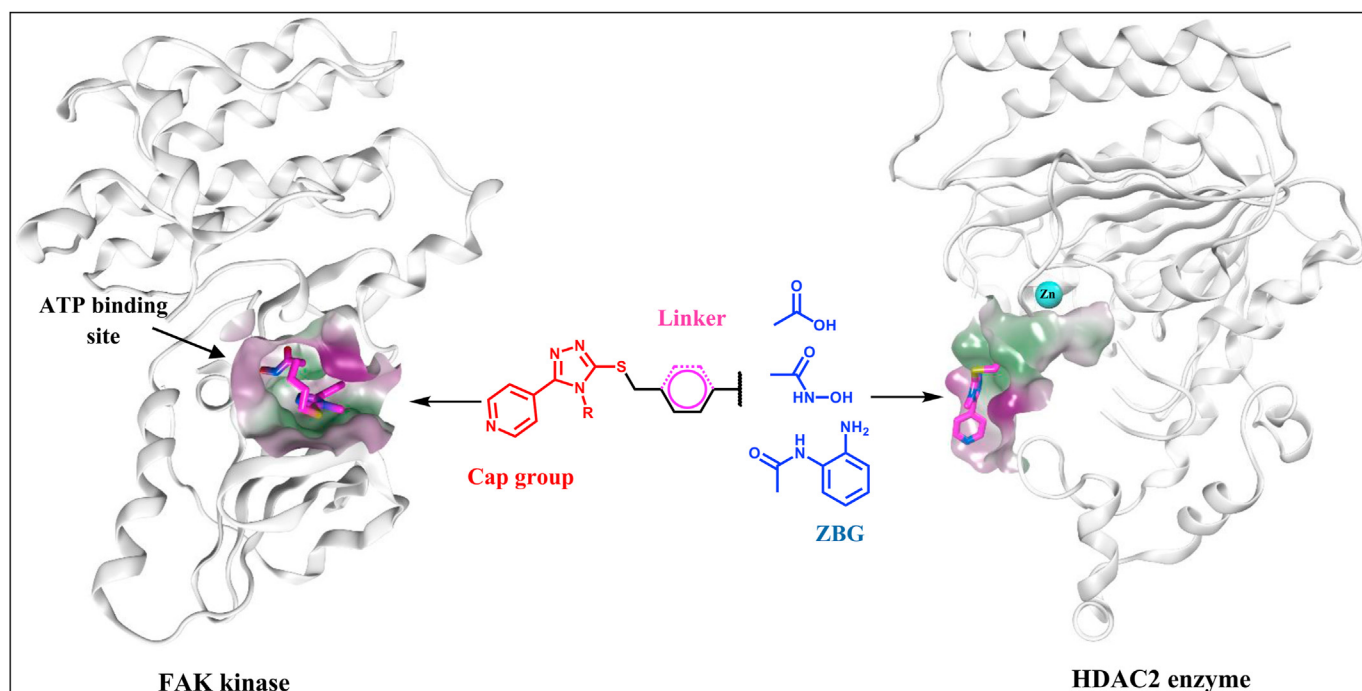

Fig. 4. Design of the dual acting 5-pyridinyl-1,2,4-triazole derivatives.

results showed that the synthesized compounds possess promising anticancer activity, especially against renal cancer, among them **5d**, **6a**, **7c**, and **11c** were the most potent.

The structure-activity relationship revealed that the N^4 methyl derivatives showed promising anticancer activity with the carboxylic acid, the hydroxamic acid, and the 2-aminobenzamide ZBGs regardless of the type of the linker (aliphatic or aromatic). Except for the aliphatic derivatives bearing hydroxamic acid or the 2-aminobenzamide motifs, the N^4 phenyl substituted derivatives showed considerable antiproliferative activity, while compounds containing N^4 ethyl did not reveal notable activity. Moreover, the

presence of 2-aminobenzamide ZBG is essential for the N^4 allyl substituted derivatives to display potent antiproliferative activity.

2.2.2. In vitro HDAC inhibition assay

The HDAC (1, 2, 3, 6, and 8) inhibitory activities of the synthesized 5-pyridinyl-1,2,4-triazole derivatives were evaluated by fluorogenic enzymatic assays compared to valproic acid (VPA), TSA, and SAHA as positive controls. The results are summarized in Table 3. The dual-acting compounds showed strong to weak inhibitory activities against HDAC (1, 2, 3, 6, and 8). The IC$_{50}$ values revealed that the synthesized compounds were more potent

Table 1
Growth inhibition (%) of 5-pyridinyl-1,2,4-triazole derivatives **5a-d**, **6a-d**, **7a-d**, **10a-d**, and **11a-d**.

Cpd	R	HCT116	HT-29	K562	KG-1	A-498	Caki-1
5a	Methyl	20.12	34.19	18.25	55.94	45.28	34.69
5b	Ethyl	2.15	9.34	1.58	7.34	4.13	1.06
5c	Allyl	16.25	21.25	2.95	15.36	11.36	3.25
5d	Phenyl	78.51	80.48	72.73	67.19	89.34	94.8
6a	Methyl	60.14	68.34	40.12	84.17	99.81	98.14
6b	Ethyl	21.35	9.15	4.09	17.28	5.26	2.06
6c	Allyl	2.07	41.28	16.12	5.91	11.06	8.62
6d	Phenyl	3.56	1.09	8.34	11.25	4.36	7.68
7a	Methyl	54.36	71.36	44.94	63.15	45.34	50.82
7b	Ethyl	6.35	18.64	10.36	3.64	22.34	8.91
7c	Allyl	90.49	95.48	86.37	94.18	97.67	95.38
7d	Phenyl	24.51	9.64	15.14	34.91	4.67	1.65
10a	Methyl	32.36	54	16	72.08	31.84	48.19
10b	Ethyl	1.03	8.34	14.29	7.12	1.67	9.34
10c	Allyl	44.13	25.34	36.08	61.95	44.85	50.21
10d	Phenyl	29.34	42.67	39.13	50.94	56.91	61.85
11a	Methyl	18.03	47.91	54.09	24.11	55.61	75.16
11b	Ethyl	1.85	1.22	8.67	5.91	1.18	12.09
11c	Allyl	80.46	70.67	90.14	87.46	97.08	94.81
11d	Phenyl	45.36	37.92	19.6	47.34	72.54	66.94

toward HDAC2 (IC$_{50}$ range: 0.09–40.07 μM) than HDAC1 (IC$_{50}$ range: 8.93->300 μM), HDAC3 (IC$_{50}$ range: 3.41->300 μM), HDAC6 (IC$_{50}$ range: 6.93->300 μM), and HDAC8 (IC$_{50}$ range: 5.32->300 μM). Compounds **5d**, **6a**, **7c**, and **11c** were the most potent derivatives in this study toward HDAC2 with IC$_{50}$ values ranging from 0.09 to 1.4 μM. The other compounds showed moderate to low activities. Moreover, compound **6a** exhibited comparable HDAC2

inhibitory activity (IC$_{50}$ = 0.090 μM), compared to the reference HDAC inhibitors TSA (IC$_{50}$ = 0.035 μM) and SAHA (IC$_{50}$ = 0.096 μM), and more potent than VPA (IC$_{50}$ = 102.74 μM). Compound **6a** was found to be more selective toward HDAC2 than HDAC1 (~137 fold), HDAC3 (~38 fold), HDAC6 (~154 fold), and HDAC8 (~79 fold). These data revealed that using the 5-pyridinyl-1,2,4-triazole scaffold, the five aliphatic/aromatic carbon linker, and various ZBGs improved HDAC2 selectivity and retained the antiproliferative activity. The hydroxamic acid derivative **6a**, bearing N^4 methyl moiety with the aliphatic chain, showed the most potent HDAC2 selectivity.

2.2.3. In vitro activity against FAK kinase

As our rationale is to target HDAC2/FAK proteins, we evaluated the synthesized 5-pyridinyl-1,2,4-triazole derivatives for their activity against the FAK enzyme using Z′-LYTE® technology, which is based on FRET (Invitrogen/Life Technologies). TAE226 was used as a positive control. The results in Table 4 showed that all the synthesized compounds have FAK inhibitory activity in the nanomolar range (IC$_{50}$ range 12.59–425.31 nM). Interestingly, with agree with the HDAC2 inhibition results, **5d**, **6a**, **7c**, and **11c** were found to be the most potent anti-FAK derivatives with IC$_{50}$ values ranging from 12.6 to 36.11 nM. Compound **6a** was also the most potent derivative in this study against FAK kinase (IC$_{50}$ = 12.59 nM), and its activity was found to be comparable to the positive anti-FAK reference, TAE226 (IC$_{50}$ = 5.20 nM). Besides, the prepared compounds showed a superior selectivity on FAK when evaluated against five different tyrosine kinases (Table S1). These results revealed that the synthesized compounds have potent anti-FAK activity; among them, **6a** is the most potent.
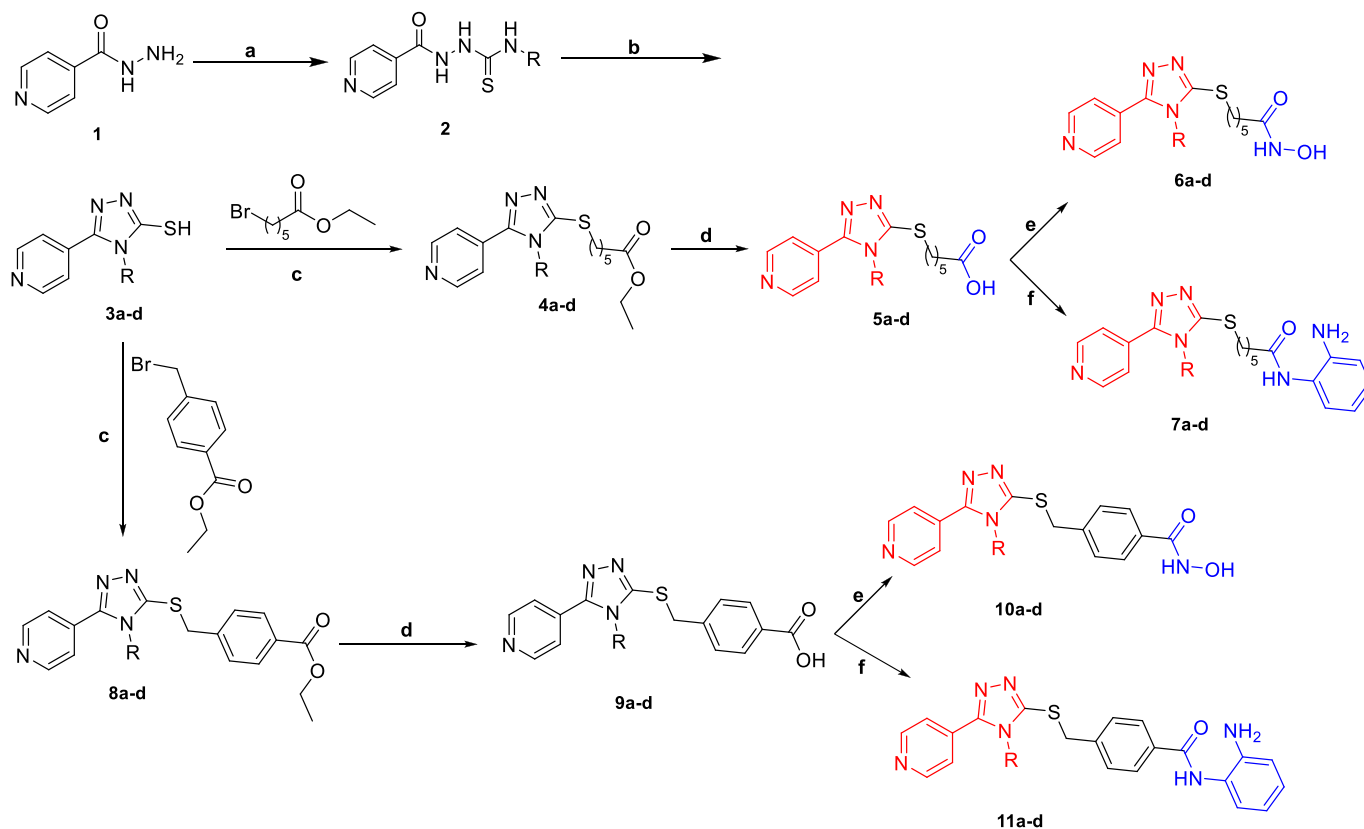
Scheme 1. Synthetic approach of the target HDAC2/FAK inhibitors
(a) RNCS (R = methyl, ethyl, allyl, or phenyl), EtOH, 70 °C, 4h; (b) (i) 2 N NaOH, (ii) HCl, 3h; (c) TEA, CH$_3$CN, 70 °C, 8h; (d) (i) LiOH, EtOH, (ii) HCl; (e) CDI, DMF, NH$_2$OH.HCl, rt, 10h; (f) CDI, DMF, 2-phenylenediamine, TFA, rt, 10h.

Table 2

Concentration producing 50% growth inhibition (IC$_{50}$, μM±SEM) of the 5-pyridinyl-1,2,4-triazole derivatives on A-498 and Caki-1 renal cancer cells.

Cpd	R	A-498	Caki-1	Cpd	R	A-498	Caki-1
5a	Methyl	12.3 ± 1.2	16.54 ± 2.49	**10a**	Methyl	20.85 ± 1.19	11.04 ± 2.15
5b	Ethyl	125.02 ± 5.23	183.34 ± 5.28	**10b**	Ethyl	203.12 ± 8.37	322.07 ± 7.61
5c	Allyl	85.34 ± 3.97	164.3 ± 3.16	**10c**	Allyl	13.25 ± 2.90	9.56 ± 2.03
5d	Phenyl	2.35 ± 0.36	1.89 ± 0.54	**10d**	Phenyl	8.76 ± 1.25	7.13 ± 1.53
6a	Methyl	0.95 ± 0.17	1.23 ± 0.35	**11a**	Methyl	9.34 ± 1.04	6.97 ± 1.58
6b	Ethyl	149.35 ± 6.49	112.03 ± 2.05	**11b**	Ethyl	124.36 ± 3.46	80.67 ± 4.93
6c	Allyl	76.36 ± 2.37	89.36 ± 2.56	**11c**	Allyl	1.42 ± 0.34	1.60 ± 0.51
6d	Phenyl	174.19 ± 5.18	92.36 ± 3.15	**11d**	Phenyl	6.19 ± 1.05	8.35 ± 1.73
7a	Methyl	13.40 ± 2.31	9.87 ± 2.49	VPA	–	2141.25 ± 40.36	2825.37 ± 22.46
7b	Ethyl	77.63 ± 3.08	98.3 ± 5.30	SAHA	–	25.08 ± 3.70	32.46 ± 2.07
7c	Allyl	1.19 ± 0.28	1.34 ± 0.34	TAE226	–	0.81 ± 0.14	1.05 ± 0.34
7d	Phenyl	341.85 ± 2.79	214.97 ± 6.01				

Table 3

HDAC inhibitory activities of 5-pyridinyl-1,2,4-triazole derivatives (IC$_{50}$, μM).

Cpd	R	HDAC1	HDAC2	HDAC3	HDAC6	HDAC8	HDAC1/HDAC2	HDAC3/HDAC2	HDAC6/HDAC2	HDAC8/HDAC2
5a	Methyl	8.93 ± 1.25	7.36 ± 0.36	84.60 ± 4.32	78.39 ± 2.15	11.95 ± 2.34	1.21	11.49	10.65	1.62
5b	Ethyl	184.36 ± 8.36	26.39 ± 2.78	156.40 ± 5.94	>300	>300	6.99	5.93	>11.37	>11.37
5c	Allyl	98.19 ± 4.37	10.12 ± 2.10	245.79 ± 4.36	>300	>300	9.70	24.29	>29.64	>29.64
5d	Phenyl	8.95 ± 2.04	0.64 ± 0.15	21.10 ± 3.17	10.89 ± 3.09	11.05 ± 4.37	13.98	32.97	17.02	17.27
6a	Methyl	12.31 ± 1.64	0.09 ± 0.02	3.41 ± 0.36	13.87 ± 1.54	7.10 ± 0.86	136.78	37.89	154.11	78.89
6b	Ethyl	76.34 ± 2.15	28.93 ± 6.35	>300	269.72 ± 11.23	>300	2.64	10.37	>9.32	10.37
6c	Allyl	>300	11.72 ± 2.30	222.41 ± 9.37	287.96 ± 2.81	11.06 ± 0.37	>25.60	18.98	24.57	0.94
6d	Phenyl	>300	32.36 ± 2.64	>300	246.13 ± 5.19	>300	>9.27	>9.27	7.61	>9.27
7a	Methyl	40.18 ± 5.02	14.36 ± 0.82	84.13 ± 1.39	14.20 ± 2.70	7.28 ± 0.49	2.80	5.86	0.99	0.51
7b	Ethyl	>300	27.13 ± 1.72	>300	>300	198.07 ± 5.32	>11.06	>11.06	>11.06	7.30
7c	Allyl	31.05 ± 3.40	1.40 ± 0.33	10.98 ± 1.03	9.79 ± 2.16	5.86 ± 0.81	22.18	7.84	6.99	4.19
7d	Phenyl	256.36 ± 11.34	38.73 ± 3.71	>300	159.46 ± 2.38	>300	6.62	>7.75	4.12	>7.75
10a	Methyl	14.2 ± 1.22	4.35 ± 0.76	56.39 ± 4.39	14.36 ± 2.32	29.76 ± 1.87	3.26	12.96	3.30	6.84
10b	Ethyl	>300	30.39 ± 2.49	174.13 ± 8.69	>300	>300	>9.87	5.73	>9.87	>9.87
10c	Allyl	19.60 ± 2.72	11.19 ± 2.18	20.70 ± 0.79	9.86 ± 1.28	28.37 ± 6.87	1.75	1.85	0.88	2.54
10d	Phenyl	15.03 ± 2.58	9.76 ± 3.90	17.96 ± 1.72	6.93 ± 0.57	12.76 ± 2.51	1.54	1.84	0.71	1.31
11a	Methyl	7.64 ± 2.79	12.97 ± 0.63	71.08 ± 2.79	11.03 ± 3.76	7.85 ± 3.19	0.59	5.48	0.85	0.61
11b	Ethyl	>300	40.07 ± 5.37	>300	>300	>300	>7.49	7.49	>7.49	>7.49
11c	Allyl	11.84 ± 1.63	1.28 ± 0.46	42.09 ± 4.72	9.96 ± 1.29	22.49 ± 3.10	9.25	32.88	7.78	17.57
11d	Phenyl	22.13 ± 3.47	8.39 ± 1.08	112.75 ± 8.91	9.63 ± 1.48	5.32 ± 0.79	2.64	13.44	1.15	0.63
VPA	–	84.21 ± 4.18	102.74 ± 3.94	134.67 ± 2.87	215.01 ± 8.56	184.30 ± 5.16	0.81	1.31	2.09	1.79
TSA	–	0.041 ± 0.012	0.035 ± 0.010	0.46 ± 0.18	0.25 ± 0.094	0.98	1.17	13.14	7.14	28.00
SAHA	–	0.062 ± 0.019	0.096 ± 0.015	0.03 ± 0.012	0.015 ± 0.008	0.42 ± 0.10	0.65	0.31	0.16	4.38

2.2.4. Inhibition of HDAC2 expression and FAK phosphorylation using western blot analysis

To further investigate the biochemical effects of the most active compounds **5d**, **6a**, **7c**, and **11c** on HDAC2 and FAK, Western blot analysis was performed on A-498 and Caki-1 cells. The obtained data were attuned to β-actin expression to exclude the variations. Among the tested compounds, **6a** maintained a promising HDAC2/FAK inhibitory activity; Treating cells with **6a** decreased the expression level of HDAC2 and FAK phosphorylation (Fig. 5). Moreover, **6a** effectively reduced their expression in A498 than Caki-1 cells. Interestingly, FAK autophosphorylation was entirely blocked by treating A-498 and Caki-1 cells with **7c**. Taken together, the mentioned results demonstrate that **6a** represses the expression of HDAC2 and the phosphorylation FAK.

2.2.5. Cell cycle and apoptosis analysis

The effect of compound **6a** on cell cycle progression in A-498 and Caki-1 cells was analyzed using flow cytometry (Fig. 6A and B). Compared to the negative control group, the percentages of cells in G2/M phase increased substantially from 18% to 65%, in A-498 cells and from 5.5% to 35%, in Caki-1 cells. Also, the percentage of cells in G1 phase decreased after treatment with 1.5 μM of compound **6a** for 24 h, whereas the percentage of cells in S phase showed minor changes. Evidently, the derivative **6a** induced the arrest of a significant percentage of cells in G2/M phase of the cell cycle compared with the untreated cells. To explore whether the antiproliferative activity of the synthesized compounds toward renal cancer cells (A-498 and Caki-1) was accompanied by an increase in cancer cell apoptosis, the apoptosis of A-498 and Caki-1 cells treated with different concentrations of the most potent derivative **6a** (0.75 and 1.5 μM) was analyzed using Annexin V-FITC/PI assay. As illustrated in Fig. 6C and D, it shows that 50% of A-498 cells experienced apoptosis with **6a** treatment. Moreover, **6a** induced apoptosis in Caki-1 cells to reach 37% at 1.5 μM concentration.

Table 4

The FAK enzymatic activities of the 5-pyridinyl-1,2,4-triazole derivatives (IC$_{50}$, nM).

Cpd	R	IC$_{50}$ (nM)	Cpd	R	IC$_{50}$ (nM)
5a	Methyl	50.30 ± 2.14	**7d**	Phenyl	312.11 ± 3.05
5b	Ethyl	76.21 ± 5.36	**10a**	Methyl	64.35 ± 3.75
5c	Allyl	124.30 ± 8.71	**10b**	Ethyl	254.36 ± 5.28
5d	Phenyl	36.11 ± 1.09	**10c**	Allyl	46.12 ± 2.01
6a	Methyl	12.59 ± 1.41	**10d**	Phenyl	22.40 ± 1.39
6b	Ethyl	124.03 ± 5.79	**11a**	Methyl	53.12 ± 2.91
6c	Allyl	189.36 ± 6.42	**11b**	Ethyl	112.15 ± 5.30
6d	Phenyl	425.31 ± 9.54	**11c**	Allyl	15.60 ± 3.10
7a	Methyl	57.08 ± 3.04	**11d**	Phenyl	96.12 ± 5.84
7b	Ethyl	140.64 ± 6.18	TAE226	–	5.20 ± 0.94
7c	Allyl	14.30 ± 2.73			

2.2.6. Relevance of Akt activation in HDAC inhibitor, VPA, resistance and relevance of Akt suppression in dual HDAC and FAK inhibitors sensitivity

It was reported that the chronic the HDAC inhibitor, VPA, application causes resistance *via* activation of Akt, and it is well known that FAK plays a vital role in Akt activation. Therefore, FAK inhibition will overcome HDAC inhibitors' resistance *via* Akt activation suppression. We hypothesized that the synthesized dual HDAC2/FAK inhibitors might not allow resistance development upon prolonged use due to their ability to inhibit FAK and eventually suppress Akt activation. To test our hypothesis, A-498 cells were exposed to a fixed dose of either the synthesized dual HDAC2/FAK inhibitor **6a** or the HDAC inhibitor VPA for 12 weeks, and then the antiproliferative activities of **6a** and VPA were reassessed. There was no marked change in the IC_{50} on **6a**-pretreated A-498 cells ($IC_{50} = 1$ μM) in comparison with its IC_{50} in cells that were not pretreated with **6a** (0.95 μM, Fig. 7A). In contrast, a significant change in the IC_{50} (from 2.1 mM to 3.09 mM) was observed in cells that were not pretreated with VPA compared to pretreated cells (Fig. 7B). These data suggest that there is no observed resistance to **6a** in A-498 cells, but the cells got resistant to VPA. Previously, researchers showed that the reason for the VPA resistance is due to Akt activation after chronic use. To elucidate the molecular mechanism by which VPA acquires resistance and **6a** maintains its sensitivity, we examined HDAC2, p-FAK, FAK, p-Akt, and Akt proteins expression in A-498 cells pretreated or not pretreated with either **6a** or VPA for 12 weeks for 24 h. The obtained results revealed that **6a** showed a steady ability to inhibit HDAC2 expression and FAK phosphorylation for 12 weeks in both pretreated and not pretreated A-498 cells (Fig. 7C). Also, it showed a substantial ability to inhibit Akt phosphorylation in both pretreated and not pretreated experiments. On the contrary, the VPA-pretreated cells showed a drastic overexpression of Akt phosphorylation, accompanied by a lower ability to inhibit HDAC2 expression (Fig. 7D).

2.2.7. Compound 6a inhibits STAT3 activation

The transcription factor, STAT3, plays a fundamental role in carcinogenesis, and is promising target for anticancer drug development. Moreover, the activation of STAT3 is well correlated with the overexpression of HDAC and FAK [17,18]. To elucidate whether the anticancer effects of **6a** are associated with STAT3 inhibition, this study examined the expression of STAT3. Western blotting results in Fig. 8 showed the ability of **6a** to inhibit the STAT3 activation.

2.2.8. Activation of different caspases as apoptotic executioners

Activation of caspases (2, 3, 6, 7, 8, 9, and 10) is a prominent biochemical feature representing a central executioner of apoptosis mediated by various inducers. HDAC/FAK inhibition is associated with caspase-8, caspase-9, and caspase-3 activation [40,41]. Consequently, Compound **6a** was evaluated for its potential to

induce caspase-8, caspase-9, and caspase-3 activation in A-498 cell line (Table 5). The exposure of A-498 cells to **6a** revealed a substantial activation of caspase-8 and caspase-9 by 3 and 6-folds, respectively, compared to the negative control. Interestingly, **6a** activated caspase-3 dramatically by 11-folds in A-498 cells compared to the negative control. The caspases activation could explain the ability of the synthesized compounds to induce apoptosis via caspases pathways.

2.3. Docking study

To further investigate the broad antiproliferative activity of the hydroxamic acid derivative **6a**, molecular docking was performed on the crystallographic structures of HDAC2 (PDB code: 4LXZ) and FAK (PDB code: 2JKK), obtained from the Protein Data Bank (PDB) [42,43]. The 4LXZ entry was selected for HDAC2 because the protein is bound with the potent HDAC inhibitor SAHA in this structure. Likewise, the entry 2JKK was chosen for FAK because, in this case, the protein is complexed with the potent FAK inhibitor TAE226. Compound **6a** showed several significant interactions within HDAC2 active site, which may account for its potent inhibitory activity (Table 6). The hydroxamic acid moiety (ZBG) exhibited a co-ordinate bond with Zn atom and formed three hydrogen bonds with the key amino acids His145, Gly154, and Tyr308 (Fig. 9A). Besides, Leu276 showed a hydrogen bonding interaction with the *N*2 of the 1,2,4-triazole ring. Moreover, the hydrophobic amino acids Phe155, Phe210, and His183 showed π … H interactions with the aliphatic linker (methylene groups). The 1,2,4-triazole and pyridine rings formed two π … H contacts with Phe155 and His33 amino acids, respectively. Additionally, **6a** occupied the ATP binding site of FAK kinase perfectly and formed approximately the same interactions as TAE226 (Table 7). Favorably, the nitrogen in the pyridinyl moiety formed a hydrogen bond with Cys502 of the kinase hinge (Fig. 9B). The pyridine ring formed four π … H contacts with Ile428, Cys502, Leu501, and Leu553 amino acids. The sulfur atom made a hydrogen bond with Gly429 amino acid, and the 5-pyridinyl-1,2,4-triazole moiety pointed toward the gatekeeper residue Met499. Moreover, the carbonyl oxygen formed two hydrogen bonds with Gly563 and Asn551 amino acids. The hydroxamic acid moiety was also in higher proximity to Asp564 of the DFG motif of the activation loop.

3. Conclusion

HDAC inhibitors comprise a diverse library of scaffolds that is amenable to several modifications, allowing the hybridization approach. FAK inhibition is emerging as a promising approach to treat cancer and prevent metastasis. The major discovery we report here is the rationale design and validation of a series of novel first-in-class dual HDAC2/FAK inhibitors containing 5-pyridinyl-1,2,4-triazole. The structural features of the HDAC scaffold included a surface recognition group (pyridine moiety), a linker (aliphatic or aromatic five-carbon linker), and a ZBG (carboxylic, hydroxamic acids, and 2-aminobenzamide). The prepared compounds showed remarkable HDAC2 selectivity among five tested HDAC isoforms. Compounds **5d**, **6a**, **7c**, and **11c** showed significant dual inhibition of HDAC2 and FAK enzymes. The same compounds revealed a promising ability to inhibit six different cancer cells (HCT116, HT-29, K562, KG-1, A-498, and Caki-1). The hydroxamic acid derivative **6a** displayed the most potent HDAC2/FAK inhibitory activity and, subsequently, the highest antiproliferative activity. Unlike VPA, Western blot analysis revealed that **6a** pretreated A-498 cells did not result in Akt activation, thus decreased the HDAC inhibition related side effects. Besides, **6a** inhibited STAT3 phosphorylation, induced cellular apoptosis, and arrested the cell cycle. It activated
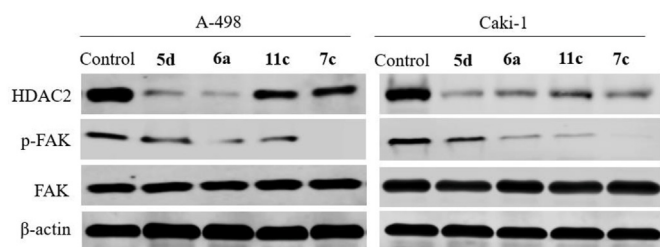

Fig. 5. Western blot analysis of HDAC2, FAK, and total FAK in A-498 and Caki-1 cells treated with **5d**, **6a**, **7c**, and **11c** (β-actin was used as loading control).
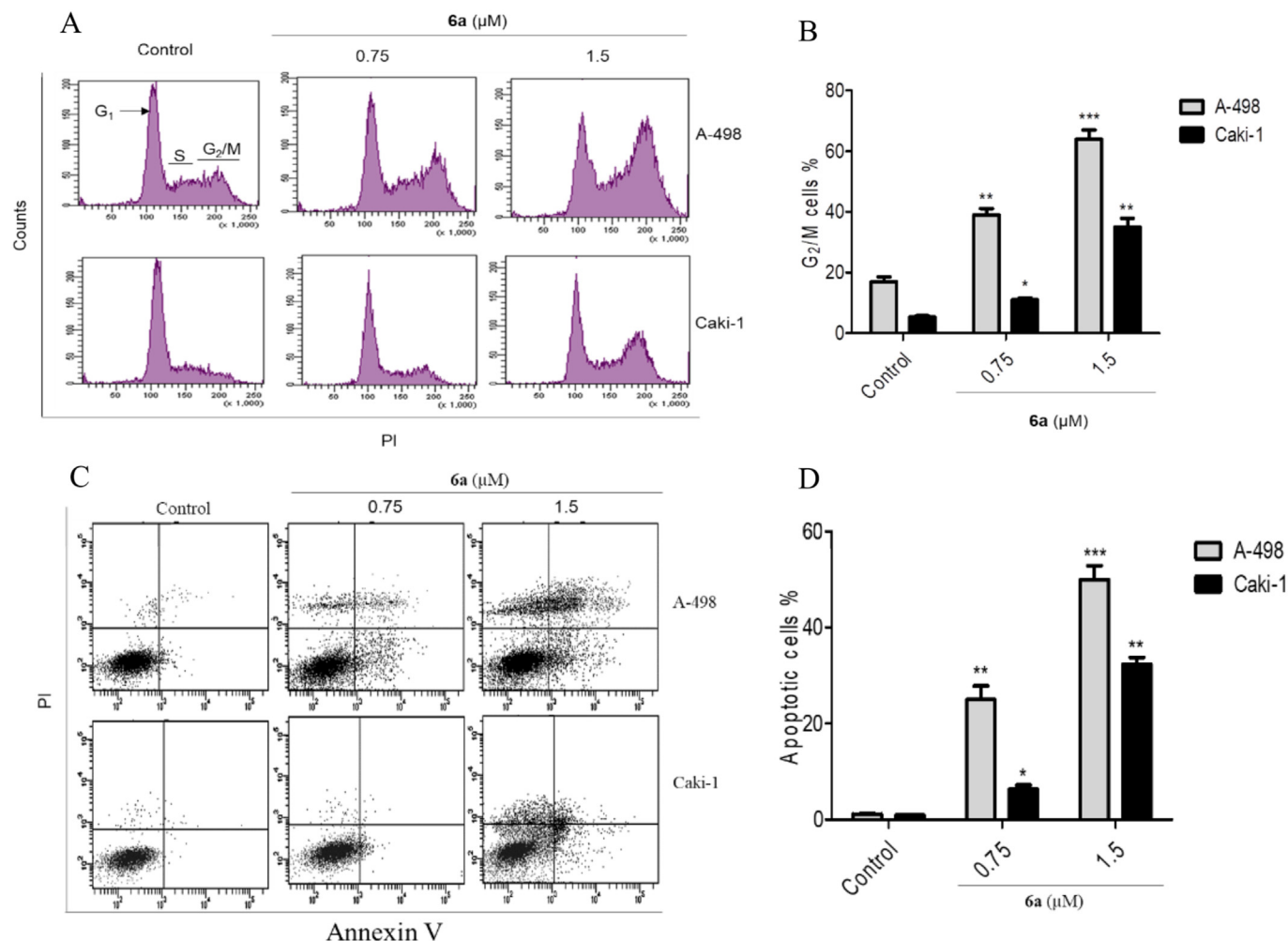
Fig. 6. Cell cycle analysis and Flow cytometry of A-498 and Caki-1 cells at 0.75 and 1.5 μM concentrations. A) **6a** induces G2/M arrest and decreases G1 phase concomitantly; B) Increasing the percentage of cells in G2/M phase parallel to increasing **6a** concentration; C) **6a** induces apoptosis in A-498 and Caki-1 cells; D) Percentage of apoptotic cells triggered by **6a** administration.

caspase-8, caspase-9, and caspase-3. Compound **6a** occupied HDAC2 and FAK active sites and formed all the essential interactions. These results suggest that developing a compound that could target HDAC/FAK proteins is considered a promising approach that warrants further investigation in animal models.

4. Experimental methods

4.1. Chemistry

All chemicals were purchased from Aldrich, Merck, Alpha Aesar, and El-Nasr pharmaceutical chemicals companies, and were used without further purification. Reactions were monitored by thin layer chromatography (TLC), using Merck9385 pre-coated aluminum plate silica gel (Kieselgel 60) 5 × 20 cm plates with a layer thickness of 0.2 mm. The spots were detected by exposure to UV-lamp at 254 nm. Melting points were determined on Stuart electro thermal melting point apparatus and were uncorrected. NMR spectra were carried out using a Bruker Avance 400 MHz ^{1}HNMR spectrometer and 100 MHz ^{13}CNMR spectrometer (Beni Sweif, Egypt), using TMS as internal reference. Chemical shifts (δ values are given in parts per million (ppm) relative to TMS DMSO-d_6 (2.5 and 3.5 ppm for ^{1}HNMR and 76.9 ppm for ^{13}CNMR) and coupling constants (J) in Hertz. Splitting patterns are designated as follows: s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet.

Elemental analyses were recorded on Shimadzu GC/MS-QP5050A, Regional center for Mycology and Biotechnology, Al-Azhar University, Cairo, Egypt. Mass spectra were recorded on Advion compact mass spectrometer (CMS) and reported as mass/charge (m/z), [ESI-H] samples were dissolved in methanol + water + TEA and [ESI + Na] samples were dissolved in methanol + water + sodium acetate, Nawah scientific center for research, Almokattam, Cairo, Egypt.

4.1.1. General procedure for the synthesis of the compounds 3a-d

The 1,2,4-triazole derivatives were prepared according to the general procedure in literatures [35]. Equimolar quantities of the isonicotinic acid hydrazide **1** (0.01 mol) and the appropriate isothiocyanate **2** (0.01 mol) were heated under reflux in absolute ethanol for 4 h. The solvent was evaporated under vacuum and 100 mL of 2 N NaOH solution was added. The solution was refluxed for 3 h, then, the reaction mixture was cooled and acidified with conc. HCl. The formed precipitate was filtered off and recrystallized from ethanol to afford white solid with 92% yield.

4.1.2. General procedure for the synthesis of the compounds 5a-d and 9a-d

The prepared 1,2,4-triazoles **3a-d** were reacted in equimolar quantities (1.4 mmol) with the substituted esters (1.4 mmol), in presence of triethylamine (1.7 mmol) as a base and CH₃CN as a
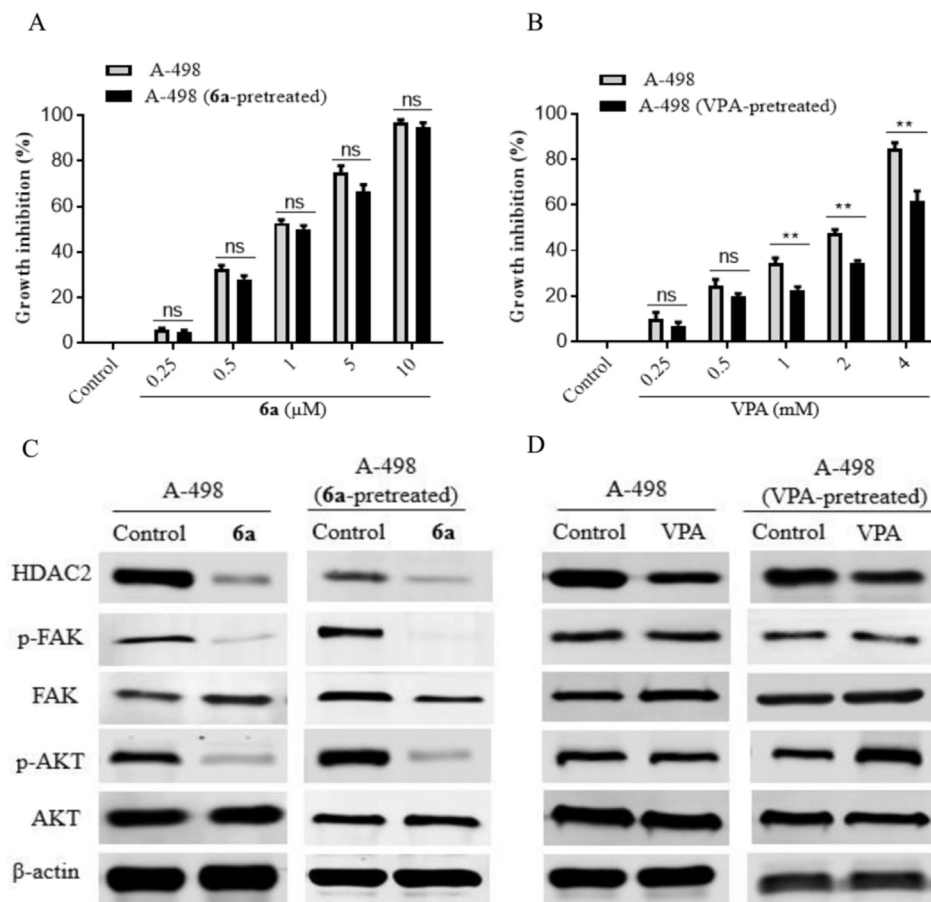
Fig. 7. Effect of the hydroxamic acid derivative **6a** on the pretreated A-498 cells. A) GI% of pretreated and not pretreated A-498 cells with **6a**; B) GI% of pretreated and not pretreated A-498 cells with VPA; C) pretreatment of A-498 cells with **6a** causes Akt activation suppression; D) pretreatment of A-498 cells with VPA causes significant Akt activation.
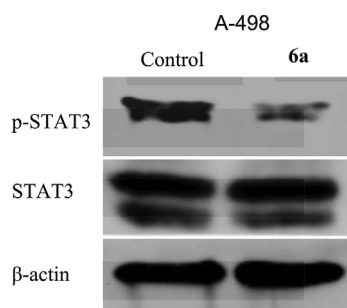

Fig. 8. Effect of the hydroxamic acid derivative **6a** on STAT3 activation. Treatment of A-498 cells with **6a** causes STAT3 activation suppression.

Table 5
Effect of compound **6a** on caspase-8, caspase-9, and caspase-3 activation.

Compound	Caspase-8	Caspase-9	Caspase-3 concentration, pg/ml
	Concentration, ng/ml		
6a	0.888	28.47	515.40
Control	0.274	4.53	45.57

solvent to yield the ester derivatives **4a-d** and **8a-d**. The corresponding carboxylic acid derivatives **5a-d** and **9a-d** (1.3 mmol), were obtained through the hydrolysis of **4a-d** and **8a-d** with 20 equivalents of LiOH solution, followed by acidification with dil. HCl.

The produced precipitate was recrystallized with aqueous ethanol with 90−95% yield.

4.1.2.1. 6-((4-methyl-5-(pyridin-4-yl)-4H-1,2,4-triazol-3-yl)thio)hexanoic acid 5a. White powder (0.38 g, 92% yield); mp 222−224 °C; ^{1}HNMR (400 MHz, DMSO-d_6) δ (ppm): 1.35−1.39 (m, 2H, C$\underline{H}_2$ aliphatic), 1.45−1.51 (m, 2H, C$\underline{H}_2$ aliphatic), 1.63−1.69 (m, 2H, C$\underline{H}_2$ aliphatic), 2.17 (t, 2H, J=7.21 Hz, C$\underline{H}_2$ aliphatic), 3.14 (t, 2H, J=7.21 Hz, S−C$\underline{H}_2$), 3.63 (s, 3H, N−C$\underline{H}_3$), 7.71 (d, 2H, J=4.81 Hz, Ar−H), 8.72 (d, 2H, J=5.61 Hz, Ar−H), 11.98 (s, 1H, COO$\underline{H}$); ^{13}CNMR (100 MHz, DMSO-d_6) δ (ppm): 24.49, 27.98, 29.40, 32.44, 32.99, 34.09, 122.76, 134.99, 150.86, 152.74, 153.71, 174.98 (C=O); Anal. Calcd for $C_{14}H_{18}N_4O_2S$: C, 54.88; H, 5.92; N,18.29; S, 10.46. Found: C, 55.12; H, 6.16; N, 18.40; S, 10.53; ESI-MS: [M − H]⁻, 305.3, found: 305.2.

4.1.2.2. 6-((4-ethyl-5-(pyridin-4-yl)-4H-1,2,4-triazol-3-yl)thio)hexanoic acid 5b. White powder (0.39 g, 93% yield); mp 205−207 °C; ^{1}HNMR (400 MHz, DMSO-d_6) δ (ppm): 1.19 (t, 3H, J=7.21 Hz, N−CH$_2$−C$\underline{H}_3$), 1.34−1.39 (m, 2H, C$\underline{H}_2$ aliphatic), 1.45−1.51 (m, 2H, C$\underline{H}_2$ aliphatic), 1.65−1.71 (m, 2H, C$\underline{H}_2$ aliphatic), 2.17 (t, 2H, J=7.21 Hz, C$\underline{H}_2$ aliphatic), 3.19 (t, 2H, J=7.21 Hz, S−C$\underline{H}_2$), 4.02 (q, 2H, J=7.21 Hz, N−C$\underline{H}_2$−CH$_3$), 7.66 (d, 2H, J=5.61 Hz, Ar−H), 8.73 (d, 2H, J=5.61 Hz, Ar−H), 12.03 (s, 1H, COO$\underline{H}$); ^{13}CNMR (100 MHz, DMSO-d_6) δ (ppm): 15.50, 24.49, 28.01, 29.40, 32.94, 34.02, 122.83, 135.14, 151.01, 152.18, 153.12, 174.97 (C=O); Anal. Calcd for $C_{15}H_{20}N_4O_2S$: C, 56.23; H, 6.29; N, 17.49; S, 10.01. Found: C, 56.45; H, 6.38; N, 17.71; S, 9.87; ESI-MS: [M − H]⁻, 319.4, found: 319.0.

Table 6
Energy scores (kcal/mol) and binding features for the HDAC2/FAK inhibitor **6a** compared to SAHA, within HDAC2 binding site.

Compound	Energy score (S) (kcal/mol)	Ligand-receptor interactions		
		Residue	Type	Length (Å)
SAHA	−9.43	Zn	Coordinate bond	2.56
		His145	Hydrogen bond	2.65
		Phe155	π … H	3.82
		His183	π … H	4.06
		Phe210	π … H	4.44
6a	−8.82	Zn	Coordinate bond	2.69
		His145	Hydrogen bond	2.70
		Gly154	Hydrogen bond	1.97
		Tyr308	Hydrogen bond	1.78
		Leu276	Hydrogen bond	3.86
		His33	π … H	4.91
		His183	π … H	4.20
		Phe155	π … H	4.0
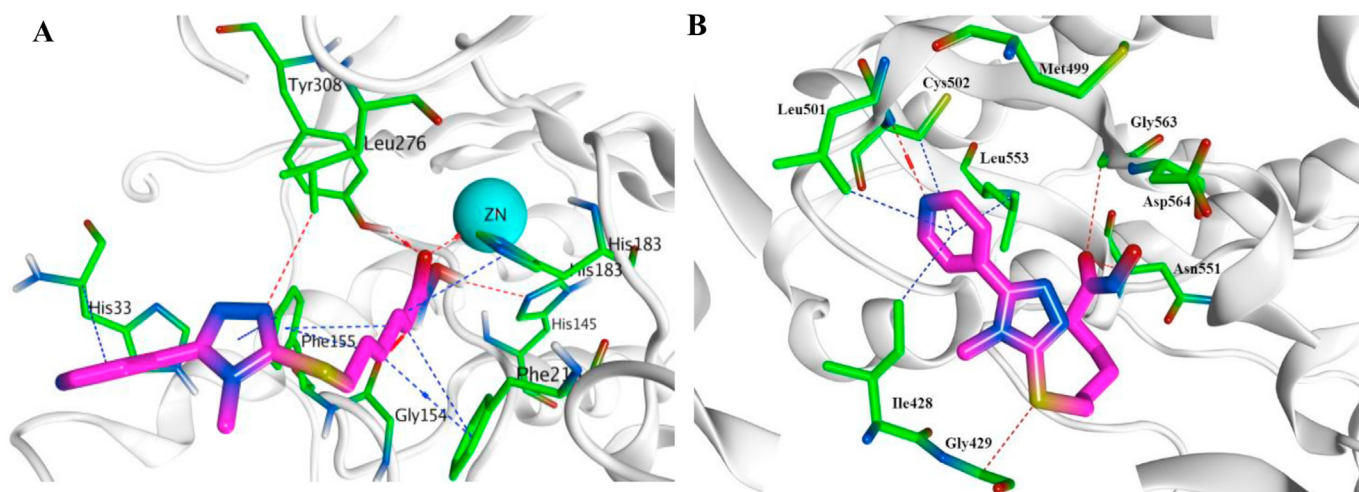
Fig. 9. Presumptive binding modes of the compound **6a** (magenta). A) Docking pose of **6a** within HDAC2 active site (PDB code: 4LXZ); B) Docking pose of **6a** within the ATP binding site of FAK (PDB code: 2JKK).

Table 7
Energy scores (kcal/mol) and binding features for the HDAC2/FAK inhibitor **6a** compared to TAE226, within the ATP binding site of FAK.

Compound	Energy score (S) (kcal/mol)	Ligand-receptor interactions		
		Residue	Type	Length (Å)
TAE226	−9.51	Cys502	Hydrogen bond Hydrogen bond	2.99
		Glu500		3.24
		Ile428	Hydrogen bond	3.83
		Asp564	Hydrogen bond	3.00
		Leu501	π … H	4.45
		Leu553	π … H	3.33
6a	−8.72	Cys502	Hydrogen bond	3.27
		Gly429	Hydrogen bond	3.69
		Ile428	Hydrogen bond	4.08
		Asn551	Hydrogen bond	3.24
		Gly563	Hydrogen bond	3.74
		Cys502	π … H	4.80
		Ile428	π … H	4.10
		Leu501	π … H	4.48
		Leu553	π … H	3.46

4.1.2.3. 6-((4-allyl-5-(pyridin-4-yl)-4H-1,2,4-triazol-3-yl)thio)hex-anoic acid 5c. Off white powder (0.41 g, 95% yield); mp 171−173 °C; ^{1}HNMR (400 MHz, DMSO-d_6) δ (ppm): 1.30−1.35 (m, 2H, C$\underline{H}_2$ aliphatic), 1.43−1.48 (m, 2H, C$\underline{H}_2$ aliphatic), 1.61−1.66 (m, 2H, C$\underline{H}_2$ aliphatic), 2.14 (t, 2H, J=6.41 Hz, C$\underline{H}_2$ aliphatic), 3.15 (t, 2H, J=7.21 Hz, S−C$\underline{H}_2$), 4.81 (d, 2H, J_{cis}=4.81 Hz, N−C$\underline{H}_2$−CH=CH$_2$), 4.73 (d, 1H, J_{cis}=16.83 Hz, −CH=C$\underline{H}_2$), 5.17 (d, 1H, J_{trans}=10.42 Hz, −CH=C$\underline{H}_2$), 5.88−5.97 (m, 1H, CH$_2$−C$\underline{H}$=CH$_2$), 7.61 (d, 2H,

J=5.61 Hz, Ar−H), 8.69 (d, 2H, J=5.61 Hz, Ar−H); ^{13}CNMR (100 MHz, DMSO-d_6) δ (ppm): 24.49, 27.98, 29.36, 33.10, 34.03, 47.11, 117.61, 122.52, 132.80, 134.87, 150.96, 152.97, 153.49, 174.94 (C=O); Anal. Calcd for $C_{16}H_{20}N_4O_2S$: C, 57.81; H, 6.06; N, 16.85; S, 9.64. Found: C, 58.04; H, 6.21; N, 16.97; S, 9.75; ESI-MS: [M − H]$^-$, 331.4, found: 331.0.

4.1.2.4. 6-((4-phenyl-5-(pyridin-4-yl)-4H-1,2,4-triazol-3-yl)thio) hexanoic acid 5d.

White crystals (0.43 g, 90% yield); mp 180−182 °C; ^{1}HNMR (400 MHz, DMSO-d_6) δ (ppm): 1.26−1.32 (m, 2H, CH$_2$ aliphatic), 1.42−1.48 (m, 2H, CH$_2$ aliphatic), 1.61−1.67 (m, 2H, CH$_2$ aliphatic), 2.15 (t, 2H, J=7.21 Hz, CH$_2$ aliphatic), 3.12 (t, 2H, J=7.21 Hz, S−CH$_2$), 7.28 (d, 2H, J=4.81 Hz, Ar−H), 7.42−7.55 (m, 2H, Ar−H), 7.52−7.56 (m, 3H, Ar−H), 8.5 (d, 2H, J=5.61 Hz, Ar−H), 12.01 (s, 1H, COOH); ^{13}CNMR (100 MHz, DMSO-d_6) δ (ppm): 24.47, 28.01, 29.23, 32.38, 34.00, 122.05, 128.20, 130.68, 130.97, 134.05, 134.47, 150.61, 152.70, 153.91, 174.97 (C=O); Anal. Calcd for $C_{19}H_{20}N_4O_2S$: C, 61.94; H, 5.47; N, 15.21; S, 8.70. Found: C, 61.72; H, 5.70; N, 15.44; S, 8.92; ESI-MS: [M − H]$^-$, 367.4, found: 367.2.

4.1.3. General procedure for synthesis of 5-pyridinyl-4H-1,2,4-triazol-3-thiol derivatives 6a-d, 7a-d, 10a-d, and 11a-d

The corresponding carboxylic acid derivatives **5a-d** and **9a-d** (1.24 mmol) were treated with CDI (1.6 mmol) in dimethylformamide or acetonitrile at 40 °C. The resulting imidazolide derivative was then reacted with hydroxylamine hydrochloride (1.6 mmol) to yield the corresponding hydroxamic acid derivatives **6a-d** and **10a-d**, while treating the imidazolide derivatives with 2-phenylenediamine (1.6 mmol) in presence of TFA (0.13 g, 1.09 mmol) yielded the corresponding 2-aminobenzamide derivatives. After reaction completion (4−10 h), the resulting mixture was poured on dist. H$_2$O. The formed precipitate was dried and recrystallized with aq. Methanol (Compounds **7a-c** were purified with flash column using CH$_2$Cl$_2$:MeOH (9:1) as an eluent.

4.1.3.1. N-hydroxy-6-((4-methyl-5-(pyridin-4-yl)-4H-1,2,4-triazol-3-yl)thio)hexanamide 6a.

White powder (0.35 g, 88% yield); mp 141−143 °C; ^{1}HNMR (400 MHz, DMSO-d_6) δ (ppm): 1.34−1.41 (m, 2H, CH$_2$ aliphatic), 1.48−1.56 (m, 2H, CH$_2$ aliphatic), 1.66−1.73 (m, 2H, CH$_2$ aliphatic), 1.96 (t, 2H, J=7.27 Hz, CH$_2$ aliphatic), 3.19 (t, 2H, J=7.21 Hz, S−CH$_2$), 3.68 (s, 3H, N−CH$_3$), 7.76 (d, 2H, J=6.11 Hz, Ar−H), 8.77 (d, 2H, J=6.11 Hz, Ar−H), 10.38 (s, 1H, OH); ^{13}CNMR (100 MHz, DMSO-d_6) δ (ppm): 25.08, 27.96, 29.31, 32.38, 32.58, 32.97, 122.68, 134.92, 150.79, 152.67, 153.65, 169.45 (C=O); Anal. Calcd for $C_{14}H_{19}N_5O_2S$: C, 52.32; H, 5.96; N, 21.79; S, 9.98. Found: C, 52.49; H, 6.13; N, 22.06; S, 10.04; ESI-MS: [M+Na]$^+$, 344.4, found: 344.2.

4.1.3.2. 6-((4-ethyl-5-(pyridin-4-yl)-4H-1,2,4-triazol-3-yl)thio)-N-hydroxyhexanamide 6b.

White powder (0.36 g, 86% yield); mp 233−235 °C; ^{1}HNMR (400 MHz, DMSO-d_6) δ (ppm): 1.24 (t, 3H, J=7.21 Hz, N−CH$_2$−CH$_3$), 1.36−1.42 (m, 2H, CH$_2$ aliphatic), 1.49−1.57 (m, 2H, CH$_2$ aliphatic), 1.68−1.76 (m, 2H, CH$_2$ aliphatic), 1.96 (t, 2H, J=7.27 Hz, CH$_2$ aliphatic), 3.23 (t, 2H, J=7.27 Hz, S−CH$_2$), 4.07 (q, 2H, J=7.21 Hz, N−CH$_2$−CH$_3$), 7.71 (d, 2H, J=6.24 Hz, Ar−H), 8.78 (d, 2H, J=6.24 Hz, Ar−H), 8.72 (s, 1H, NH), 10.38 (s, 1H, OH); ^{13}CNMR (100 MHz, DMSO-d_6) δ (ppm): 15.44, 25.09, 28.00, 29.32, 32.58, 32.91, 122.75, 135.08, 150.94, 152.12, 153.06, 169.45 (C=O); Anal. Calcd for $C_{15}H_{21}N_5O_2S$: C, 53.71; H, 6.31; N, 20.88; S, 9.56. Found: C, 53.98; H, 6.52; N, 20.74; S, 9.67; ESI-MS: [M+Na]$^+$, 358.4, found: 357.9.

4.1.3.3. 6-((4-allyl-5-(pyridin-4-yl)-4H-1,2,4-triazol-3-yl)thio)-N-hydroxyhexanamide 6c.

Off white powder (0.35 g, 82% yield); mp 143−145 °C; ^{1}HNMR (400 MHz, DMSO-d_6) δ (ppm): 1.33−1.40 (m, 2H, CH$_2$ aliphatic), 1.48−1.55 (m, 2H, CH$_2$ aliphatic), 1.65−1.73 (m, 2H, CH$_2$ aliphatic), 1.96 (t, 2H, J=7.23 Hz, CH$_2$ aliphatic), 3.20 (t, 2H, J=7.28 Hz, S−CH$_2$), 4.72 (d, 2H, J_{cis}=4.40 Hz, N−CH$_2$−CH=CH$_2$), 4.79 (d, 1H, J_{cis}=17.45 Hz, −CH=CH$_2$), 5.23 (d, 1H, J_{trans}=10.46 Hz, −CH=CH$_2$), 5.97−6.02 (m, 1H, CH$_2$−CH=CH$_2$), 7.68 (d, 2H, J=5.99 Hz, Ar−H), 8.75 (d, 2H, J=5.99 Hz, Ar−H), 8.72 (s, 1H, NH), 10.38 (s, 1H, OH); ^{13}CNMR (100 MHz, DMSO-d_6) δ (ppm): 25.08, 27.97, 29.28, 32.57, 33.05, 47.05, 117.55, 121.67, 122.46, 132.72, 134.79, 150.77, 150.89, 152.91, 153.43, 169.45 (C=O); Anal. Calcd for $C_{16}H_{21}N_5O_2S$: C, 55.31; H, 6.09; N, 20.16; S9.23. Found: C, 55.60; H, 6.23; N, 20.45; S, 9.26; ESI-MS: [M+Na]$^+$, 370.1, found: 370.1.

4.1.3.4. N-hydroxy-6-((4-phenyl-5-(pyridin-4-yl)-4H-1,2,4-triazol-3-yl)thio)hexanamide 6d.

White crystals (0.38 g, 85% yield); mp 163−165 °C; ^{1}HNMR (400 MHz, DMSO-d_6) δ (ppm): 1.29−1.35 (m, 2H, CH$_2$ aliphatic), 1.46−1.53 (m, 2H, CH$_2$ aliphatic), 1.64−1.72 (m, 2H, CH$_2$ aliphatic), 1.94 (t, 2H, J=7.29 Hz, CH$_2$ aliphatic), 3.17 (t, 2H, J=7.23 Hz, S−CH$_2$), 7.30 (d, 2H, J=6.24 Hz, Ar−H), 7.47−7.50 (m, 2H, Ar−H), 7.57−7.61 (m, 3H, Ar−H), 8.56 (d, 2H, J=6.24 Hz, Ar−H), 8.73 (s, 1H, NH); 10.38 (s, 1H, OH); ^{13}CNMR (100 MHz, DMSO-d_6) δ (ppm): 25.06, 28.00, 29.14, 32.35, 32.56, 121.99, 128.11, 130.62, 130.91, 133.97, 134.40, 150.53, 152.63, 153.85, 169.46 (C=O); Anal. Calcd for $C_{19}H_{21}N_5O_2S$: C, 59.51; H, 5.52; N, 18.26; S, 8.36. Found: C, 59.74; H, 5.59; N, 18.42; S, 8.45; ESI-MS: [M+Na]$^+$, 406.4, found: 405.9.

4.1.3.5. N-(2-aminophenyl)-6-((4-methyl-5-(pyridin-4-yl)-4H-1,2,4-triazol-3-yl)thio)hexanamide 7a.

Brown oil (0.40 g, 81% yield); ^{1}HNMR (400 MHz, DMSO-d_6) δ (ppm): 1.49−1.55 (m, 2H, CH$_2$ aliphatic), 1.64−1.72 (m, 2H, CH$_2$ aliphatic), 1.76−1.83 (m, 2H, CH$_2$ aliphatic), 2.38 (t, 2H, J=7.34 Hz, CH$_2$ aliphatic), 3.26 (t, 2H, J=7.21 Hz, S−CH$_2$), 3.72 (s, 3H, N−CH$_3$), 4.93 (s, 2H, NH$_2$), 6.58 (t, 1H, J=7.52 Hz, Ar−H), 6.77 (d, 1H, J=9.41 Hz, Ar−H), 6.94 (t, 1H, J=7.27 Hz, Ar−H), 7.20 (d, 1H, J=9.41 Hz, Ar−H), 7.81 (d, 2H, J=6.11 Hz, Ar−H), 8.81 (d, 2H, J=6.11 Hz, Ar−H), 9.20 (s, 1H, C=O−NH); ^{13}CNMR (100 MHz, DMSO-d_6) δ (ppm): 25.24, 28.03, 29.36, 32.37, 33.02, 36.05, 116.35, 116.64, 122.17, 122.68, 123.98, 125.79, 135.64, 142.38, 150.78, 152.69, 153.65, 171.55 (C=O); Anal. Calcd for $C_{20}H_{24}N_6OS$: C, 60.58; H, 6.10; N, 21.20; S, 8.09. Found: C, 60.41; H, 6.28; N, 20.98; S, 8.27; ESI-MS: [M+Na]$^+$, 419.5, found: 419.2.

4.1.3.6. N-(2-aminophenyl)-6-((4-ethyl-5-(pyridin-4-yl)-4H-1,2,4-triazol-3-yl)thio)hexanamide 7b.

Brown oil (0.43 g, 86% yield); ^{1}HNMR (400 MHz, DMSO-d_6) δ (ppm): 1.24 (t, 3H, J=7.27 Hz, CH$_2$−CH$_3$), 1.46−1.50 (m, 2H, CH$_2$ aliphatic), 1.60−1.68 (m, 2H, CH$_2$ aliphatic), 1.74−1.81 (m, 2H, CH$_2$ aliphatic), 2.34 (t, 2H, J=7.34 Hz, CH$_2$ aliphatic), 3.27 (t, 2H, J=7.21 Hz, S−CH$_2$), 4.07 (q, 2H, J=7.27 Hz, N−CH$_2$−CH$_3$), 4.84 (s, 2H, NH$_2$), 6.53 (t, 1H, J=7.52 Hz, Ar−H), 6.72 (d, 1H, J=9.54 Hz, Ar−H), 6.87−6.92 (m, 1H, Ar−H), 7.16 (d, 1H, J=9.54 Hz, Ar−H), 7.71 (d, 2H, J=6.11 Hz, Ar−H), 8.77 (d, 2H, J=6.11 Hz, Ar−H), 9.13 (s, 1H, C=O−NH); ^{13}CNMR (100 MHz, DMSO-d_6) δ (ppm): 15.45, 25.24, 28.06, 29.37, 32.98, 30.05, 116.33, 116.61, 117.72, 122.75, 123.99, 125.76, 126.18, 135.08, 142.36, 150.94, 152.12, 153.06, 171.51 (C=O); Anal. Calcd for $C_{21}H_{26}N_6OS$: C, 61.44; H, 6.38; N, 20.47; S, 7.81. Found: C, 61.68; H, 6.62; N, 20.71; S, 7.94; ESI-MS: [M+Na]$^+$, 433.5, found: 433.3.

4.1.3.7. 6-((4-allyl-5-(pyridin-4-yl)-4H-1,2,4-triazol-3-yl)thio)-N-(2-aminophenyl)hexanamide 7c.

Brown oil (0.42 g, 92% yield); ^{1}HNMR (400 MHz, DMSO-d_6) δ (ppm): 1.44−1.50 (m, 2H, CH$_2$ aliphatic), 1.60−1.67 (m, 2H, CH$_2$ aliphatic), 1.72−1.79 (m, 2H, CH$_2$ aliphatic), 2.34 (t, 2H, J=7.34 Hz, CH$_2$ aliphatic), 3.24 (t, 2H, J=7.21 Hz, S−CH$_2$), 4.72 (d, 2H, J_{cis}=4.52 Hz, N−CH$_2$−CH=CH$_2$), 4.78−4.88 (m, 3H, J_{cis}=17.24 Hz, −CH=CH$_2$ and NH$_2$), 5.24 (d, 1H, J_{trans}=10.39 Hz, −CH=CH$_2$), 5.94−6.03 (m, 1H, CH$_2$−CH=CH$_2$), 6.54 (t, 1H,

J=7.46 Hz, Ar–H), 6.72 (d, 1H, J=9.54 Hz, Ar–H), 6.90 (t, 1H J=7.40 Hz, Ar–H), 7.16 (d, 1H, J=7.95 Hz, Ar–H), 7.69 (d, 2H, J=6.24 Hz, Ar–H), 8.76 (d, 2H, J=6.24 Hz, Ar–H), 9.12 (s, 1H, C= O–NH); ^{13}CNMR (100 MHz, DMSO-d_6) δ (ppm): 24.78, 27.58, 28.88, 32.67, 35.60, 46.59, 115.89, 116.18, 117.11, 121.76, 122.02, 123.44, 125.73, 132.28, 133.91, 134.12, 135.20, 141.88, 150.45, 170.79 (C=O); Anal. Calcd for $C_{22}H_{26}N_6OS$: C, 62.53; H, 6.20; N, 19.89; S, 7.59. Found: C, 62.77; H, 6.39; N, 20.06; S, 7.73; ESI-MS: [M+Na]$^+$, 445.5, found: 445.3.

4.1.3.8. N-(2-aminophenyl)-6-((4-phenyl-5-(pyridin-4-yl)-4H-1,2,4-triazol-3-yl)thio)hexanamide 7d. yellowish brown powder (0.48 g, 92% yield); mp 179−181 °C; ^{1}HNMR (400 MHz, DMSO-d_6) δ (ppm): 1.37−1.44 (m, 2H, CH$_2$ aliphatic), 1.57−1.65 (m, 2H, CH$_2$ aliphatic), 1.70−1.78 (m, 2H, CH2 aliphatic), 2.32 (t, 2H, J = 7.40 Hz, CH$_2$ aliphatic), 3.20 (t, 2H, J = 7.21 Hz, S–CH$_2$), 4.83 (s, 2H, NH$_2$), 6.53 (t, 1H, J=7.52 Hz, Ar–H), 6.72 (d, 1H, J=9.54 Hz, Ar–H), 6.89 (t, 1H J=7.58 Hz, Ar–H), 7.15 (d, 1H, J=9.54 Hz, Ar–H), 7.30 (d, 2H, J=6.24 Hz, Ar–H), 7.47−7.50 (m, 2H, Ar–H), 7.56−7.61 (m, 3H, Ar–H), 8.57 (d, 2H, J=6.24 Hz, Ar–H), 9.11 (s, 1H, C=O–NH); ^{13}CNMR (100 MHz, DMSO-d_6) δ (ppm): 25.23, 28.07, 29.21, 32.38, 36.05, 116.34, 116.63, 121.98, 123.99, 125.75, 126.17, 128.12, 130.61, 130.89, 133.99, 134.41, 142.35, 150.54, 152.63, 153.85, 171.50 (C=O); Anal. Calcd for $C_{25}H_{26}N_6OS$: C, 65.48; H, 5.71; N, 18.33; S, 6.99. Found: C, 65.74; H, 5.98; N, 18.50; S, 7.02; ESI-MS: [M+Na]$^+$, 481.5, found: 481.3.

4.1.3.9. N-hydroxy-4-(((4-methyl-5-(pyridin-4-yl)-4H-1,2,4-triazol-3-yl)thio)methyl)benzamide 10a. White powder (0.36 g, 81% yield); mp 190−192 °C; ^{1}HNMR (400 MHz, DMSO-d_6) δ (ppm): 3.64 (s, 3H, N–CH$_3$), 4.53 (s, 2H, S–CH$_2$), 7.51 (d, 2H, J=6.24 Hz, Ar–H), 7.75 (d, 2H, J=6.24 Hz, Ar–H), 7.79 (d, 2H, J=4.28 Hz, Ar–H), 8.83 (d, 2H, J=4.28 Hz, Ar–H), 9.1 (s, 1H, NH), 11.27 (s, 1H, OH); ^{13}CNMR (100 MHz, DMSO-d_6) δ (ppm): 32.38, 36.84, 122.67, 127.52, 129.94, 132.39, 135.47, 140.49, 150.83, 152.34, 155.07, 164.35 (C=O); Anal. Calcd for $C_{16}H_{15}N_5O_2S$: C, 56.29; H, 4.43; N, 20.51; S, 9.39. Found: C, 56.61; H, 4.56; N, 20.74; S, 9.51; ESI-MS: [M − H]$^-$, 340.3, found: 340.0.

4.1.3.10. 4-(((4-ethyl-5-(pyridin-4-yl)-4H-1,2,4-triazol-3-yl)thio) methyl)-N-hydroxybenzamide 10b. White powder (0.35 g, 76% yield); mp 198−200 °C; ^{1}HNMR (400 MHz, DMSO-d_6) δ (ppm): 1.17 (t, 3H, J=7.27 Hz, N–CH$_2$–CH$_3$), 4.03 (q, 2H, J=7.27 Hz, N–CH$_2$–CH$_3$), 4.58 (s, 2H, S–CH$_2$), 7.52 (d, 2H, J=8.31Hz, Ar–H), 7.73−7.76 (m, 4H, Ar–H), 8.82 (d, 2H, J=6.11 Hz, Ar–H), 9.10 (s, 1H, NH–NH), 11.26 (s, 1H, OH); ^{13}CNMR (100 MHz, DMSO-d_6) δ (ppm): 15.38, 36.89, 39.35, 122.73, 127.54, 129.47, 132.39, 135.00, 141.01, 150.97, 151.39, 153.22, 164.33 (C=O); Anal. Calcd for $C_{17}H_{17}N_5O_2S$: C, 57.45; H, 4.82; N, 19.71; S, 9.02. Found: C, 57.72; H, 5.06; N, 19.97; S, 9.14; ESI-MS: [M − H]$^-$, 354.4, found: 353.8.

4.1.3.11. 4-(((4-allyl-5-(pyridin-4-yl)-4H-1,2,4-triazol-3-yl)thio) methyl)-N-hydroxybenzamide 10c. White powder (0.41 g, 86% yield); mp 194−196 °C; ^{1}HNMR (400 MHz, DMSO-d_6) δ (ppm): 4.52 (s, 2H, S–CH$_2$), 4.64 (d, 2H, J=4.65 Hz, N–CH$_2$–CH=CH$_2$), 4.73 (d, 1H, J_{cis}=17.24 Hz, −CH=CH$_2$), 5.17 (d, 1H, J_{trans}=10.39 Hz, −CH= CH$_2$), 5.86−5.94 (m, 1H, CH$_2$–CH=CH$_2$), 7.48 (d, 2H, J=8.29 Hz, Ar–H), 7.66 (d, 2H, J=6.11 Hz, Ar–H), 7.70 (d, 2H, J=8.31 Hz, Ar–H), 8.76 (d, 2H, J=6.11 Hz, Ar–H), 9.05 (s, 1H, NH), 11.21 (s, 1H, OH); ^{13}CNMR (100 MHz, DMSO-d_6) δ (ppm): 36.83, 47.04, 117.67, 122.49, 127.52, 129.48, 132.40, 132.59, 134.69, 140.89, 150.92, 152.20, 153.59, 164.34 (C=O); Anal. Calcd for $C_{18}H_{17}N_5O_2S$: C, 58.84; H, 4.66; N, 19.06; S, 8.73. Found: C, 59.11; H, 4.83; N, 19.28; S, 8.59; ESI-MS: [M+Na]$^+$, 390.4, found: 390.2.

4.1.3.12. N-hydroxy-4-(((4-phenyl-5-(pyridin-4-yl)-4H-1,2,4-triazol-3-yl)thio)methyl)benzamide 10d. White powder (0.48 g, 92% yield); mp 145−147 °C; ^{1}HNMR (400 MHz, DMSO-d_6) δ (ppm): 4.54 (s, 2H, S–CH$_2$), 7.32−7.35 (m, 2H, Ar–H), 7.43−7.51 (m, 4H, Ar–H), 7.61−7.65 (m, 3H, Ar–H), 7.74 (d, 2H, J=5.14 Hz, Ar–H), 8.61 (d, 2H, J=6.11 Hz, Ar–H), 9.11 (s, 1H, NH), 11.27 (s, 1H, OH); ^{13}CNMR (100 MHz, DMSO-d_6) δ (ppm): 36.01, 121.98, 127.50, 128.04, 129.49, 130.61, 130.95, 132.37, 133.80, 134.31, 140.86, 150.59, 152.81, 153.22, 164.37 (C=O); Anal. Calcd for $C_{21}H_{17}N_5O_2S$: C, 62.52; H, 4.25; N, 17.36; S, 7.95. Found: C, 62.70; H, 4.42; N, 17.59; S, 8.06; ESI-MS: [M − H]$^-$, 402.4, found: 402.2.

4.1.3.13. N-(2-aminophenyl)-4-(((4-methyl-5-(pyridin-4-yl)-4H-1,2,4-triazol-3- yl)thio)methyl)benzamide 11a. Light yellow crystals (0.47 g, 91% yield); mp 244−246 °C; ^{1}HNMR (400 MHz, DMSO-d_6) δ (ppm): 3.65 (s, 3H, N–CH$_3$), 4.55 (s, 2H, S–CH$_2$), 4.94 (s, 2H, NH$_2$), 6.64 (t, 1H, J=7.52 Hz, Ar–H), 6.83 (d, 1H, J=8.07 Hz, Ar–H), 7.02 (t, 1H, J=7.85 Hz, Ar–H), 7.20 (d, 1H, J=7.83 Hz, Ar–H), 7.56 (d, 2H, J=6.24 Hz, Ar–H), 7.78 (d, 2H, J=4.28 Hz, Ar–H), 7.97 (d, 2H, J=8.31 Hz, Ar–H), 8.81 (d, 2H, J=4.28 Hz, Ar–H), 9.70 (s, 1H, C= O–NH); ^{13}CNMR (100 MHz, DMSO-d_6) δ (ppm): 32.43, 36.90, 116.57, 116.71, 122.66, 123.68, 127.02, 127.22, 128.44, 129.38, 134.20, 134.82, 141.24, 143.66, 150.83, 151.87, 153.85, 165.42 (C=O); Anal. Calcd for $C_{22}H_{20}N_6OS$: C, 63.44; H, 4.84; N, 20.18; S, 7.70. Found: C, 63.71; H, 4.95; N, 20.37; S, 7.65; ESI-MS: [M+Na]$^+$, 439.5, found: 439.2.

4.1.3.14. N-(2-aminophenyl)-4-(((4-ethyl-5-(pyridin-4-yl)-4H-1,2,4-triazol-3-yl)thio)methyl)benzamide 11b. Brown oil (0.39 g, 73% yield); ^{1}HNMR (400 MHz, DMSO-d_6) δ (ppm): 1.16 (t, 3H, J=7.34 Hz, CH$_2$–CH$_3$), 4.01 (q, 2H, J=7.34 Hz, CH$_2$–CH$_3$), 4.58 (s, 2H, S–CH$_2$), 4.93 (s, 2H, NH$_2$), 6.61 (t, 1H, J=7.70 Hz, Ar–H), 6.80 (d, 1H, J=8.19 Hz, Ar–H), 6.98 (t, 1H, J=7.70 Hz, Ar–H), 7.18 (d, 1H, J=8.19 Hz, Ar–H), 7.55 (d, 2H, J=8.07 Hz, Ar–H), 7.70 (d, 2H, J=5.50 Hz, Ar–H), 7.96 (d, 2H, J=8.31 Hz, Ar–H), 8.78 (d, 2H, J=5.50 Hz, Ar–H), 9.71 (s, 1H, C=O–NH); ^{13}CNMR (100 MHz, DMSO-d_6) δ (ppm): 15.44, 36.86, 39.34, 116.59, 116.72, 122.73, 123.70, 127.01, 127.21, 128.46, 129.37, 134.21, 134.99, 141.25, 143.65, 150.98, 151.40, 153.23, 165.42 (C=O); Anal. Calcd for $C_{23}H_{22}N_6OS$: C, 64.17; H, 5.15; N, 19.52; S, 7.45. Found: C, 64.40; H, 5.39; N, 19.78; S, 7.62; ESI-MS: [M+Na]$^+$, 453.5, found: 453.2.

4.1.3.15. 4-(((4-allyl-5-(pyridin-4-yl)-4H-1,2,4-triazol-3-yl)thio) methyl)-N-(2-aminophenyl)benzamide 11c. Yellowish brown powder (0.51 g, 93% yield); mp 182−184 °C; ^{1}HNMR (400 MHz, DMSO-d_6) δ (ppm): 4.61 (s, 2H, S–CH$_2$), 4.71 (d, 2H, J=4.52 Hz, N–CH$_2$–CH=CH$_2$), 4.79 (d, 1H, J_{cis}=17.24 Hz, −CH=CH$_2$), 4.96 (s, 2H, NH$_2$), 5.24 (d, 1H, J_{trans}=10.64 Hz, −CH=CH$_2$), 5.92−6.01 (m, 1H, CH$_2$–CH=CH$_2$), 6.65 (t, 1H, J=7.58 Hz, Ar–H), 6.83 (d, 1H, J=8.07 Hz, Ar–H), 7.03 (t, 1H, J=7.58 Hz, Ar–H), 7.21 (d, 1H, J=7.95 Hz, Ar–H), 7.59 (d, 2H, J=8.19 Hz, Ar–H), 7.71 (d, 2H, J=4.65 Hz, Ar–H), 7.99 (d, 2H, J=8.19 Hz, Ar–H), 8.80 (d, 2H, J=4.65 Hz, Ar–H), 9.71 (s, 1H, C=O–NH); ^{13}CNMR (100 MHz, DMSO-d_6) δ (ppm): 36.84, 47.07, 116.57, 116.71, 117.71, 122.49, 123.69, 127.01, 127.20, 128.43, 129.38, 132.62, 134.22, 134.69, 141.15, 143.65, 150.93, 152.20, 153.60, 165.42 (C=O); Anal. Calcd for $C_{24}H_{22}N_6OS$: C, 65.14; H, 5.01; N, 18.99; S, 7.24. Found: C, 65.38; H, 5.17; N, 19.23; S, 7.45; ESI-MS: [M+Na]$^+$, 465.5, found: 465.2.

4.1.3.16. N-(2-aminophenyl)-4-(((4-phenyl-5-(pyridin-4-yl)-4H-1,2,4-triazol-3-yl)thio)methyl)benzamide 11d. Off white powder (0.54 g, 91% yield); mp 230−232 °C; ^{1}HNMR (400 MHz, DMSO-d_6) δ (ppm): 4.58 (s, 2H, S–CH$_2$), 4.96 (s, 2H, NH$_2$), 6.65 (t, 1H, J=7.58 Hz, Ar–H), 6.83 (d, 1H, J=7.95Hz, Ar–H), 7.02 (t, 1H, J=7.58 Hz, Ar–H), 7.21 (d, 1H, J=7.82 Hz, Ar–H), 7.35 (d, 2H, J=6.24 Hz, Ar–H), 7.47 (d, 2H, J=8.19 Hz, Ar–H), 7.56−7.64 (m, 5H,

Ar—H), 7.97 (d, 2H, J=8.19 Hz, Ar—H), 8.62 (d, 2H, J=4.52 Hz, Ar—H), 9.70 (s, 1H, C=O—N$\underline{H}$); ^{13}CNMR (100 MHz, DMSO-d_6) δ (ppm): 34.07, 114.63, 114.77, 120.03, 121.77, 125.06, 125.26, 126.10, 126.46, 127.44, 128.70, 129.02, 131.88, 132.35, 139.18, 141.70, 148.64, 148.73, 150.87, 151.26, 163.50 (C=O); Anal. Calcd for $C_{27}H_{22}N_6OS$: C, 67.76; H, 4.63; N, 17.56; S, 6.70. Found: C, 68.02; H, 4.80; N, 17.82; S, 6.83; ESI-MS: [M+Na]$^+$, 501.5, found: 501.2.

4.2. Biological investigations

4.2.1. Cell culture and reagents

Six different cell lines derived from three tumor subpanels, including colon (HCT116 and HT-29), leukemia (K562, and KG-1), and renal (A-498 and Caki-1) cell lines were obtained from the American Type Culture Collection (ATCC; Manassas, VA, USA) and were cultured in their suitable media containing 10% fetal bovine serum (FBS; Sigma-Aldrich), 100 U/ml penicillin, and 100 μg/ml streptomycin (Life technologies) in a humidified atmosphere with 5% CO_2 at 37 °C. 5-pyridinyl-1,2,4-triazole derivatives were chemically synthesized (**5a-d**, **6a-d**, **7a-d**, **10a-d**, and **11a-d**). All chemicals were purchased from Sigma-Aldrich Co., LLC except TAE226 was purchased from Selleck Chemicals (Houston, TX, USA). All chemicals used in this study were of the analytical or cell-culture grade.

4.2.2. 3-(4,5-Dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) assay

The proliferation assay for six different cell lines derived from three tumor subpanels, including colon (HCT116 and HT-29), leukemia (K562, and KG-1), and renal (A-498 and Caki-1) cell lines was quantified using MTT assay. Cells were seeded as 1×10^4 cells/well and cultured overnight in a 96-well plate. The cells were treated with either 10 μM of 5-pyridinyl-1,2,4-triazole derivatives (**5a-d**, **6a-d**, **7a-d**, **10a-d**, and **11a-d**), or DMSO as a negative control. After 24 h, the cells washed with phosphate buffered saline (PBS, Invitrogen Gibco) and incubated with 20 μl of MTT solution (2 mg/ml) for 4 h at 37 °C. Then, 150 μl DMSO was used to solubilize MTT formazan crystals. Finally, the plates were shaken, and the optical density was determined at 570 nm using ELISA plate reader (Model 550, Bio-Rad, USA). At least, three independent experiments were performed. Percentage of growth inhibition was determined as (1− [OD of treated cells/OD of control cells]). On the other hand, using the MTT assay, we tested the effect of different concentrations (1, 5, 10, 25, 50, 100, 1000, 2000, and 4000 μM) of the synthesized compounds on renal cancer cells (A-498 and Caki-1), using DMSO as a negative control, whereas VPA, SAHA, and TAE226 were used as positive controls. The IC_{50} values were calculated using Prism v.8 software (GraphPad Software Inc., La Jolla, CA).

4.2.3. Fluorogenic enzymatic assays for HDAC isoforms

To determine the ability of the synthesized compounds to inhibit the activity of HDACs, we did invitro optimize fluorogenic enzymatic assays for HDAC isoforms in 96-well opaque half-area microplate (Corning). Recombinant, full-length human HDAC1, 2, 3, and 8 (BPS Bioscience, CA, USA) were diluted in assay buffer A (25 mM Tris—HCl, 137 mM NaCl, 2.7 mM KCl, 1 mM MgCl2, and 1 mg/mL BSA, pH 8.0) to give 4, 5, 1, 8.5 ng/μL stock solutions, respectively. Serial dilutions of inhibitors were made in assay buffer B (25 mM Tris—HCl, 137 mM NaCl, 2.7 mM KCl, 1 mM MgCl2, pH 8.0). Then, we mixed 10 μL of each enzyme stock to 30 μL of inhibitor and incubated for 5 min at room temperature. After incubation, 10 μL of HDAC substrate was added and the mixture incubated for 45 min at room temperature. For HDAC1, 2, 3 and 8, the reaction was quenched with trypsin and trichostatin A dissolved in assay buffer B and incubated at room temperature for

45 min. Fluorescence measurements were obtained with Synergy 4 Hybrid microplate reader (Bio-Tek) at excitation wavelength 360 nm and emission wavelength 460 nm. On the other hand, to determine the ability of the synthesized compounds to inhibit the activity of HDAC6, we used fluorogenic Assay kit (BPS Biosciences) per the manufacturer's instructions. Valproic acid, trichostatin A and SAHA were utilized as inhibitor positive controls. The experiments were repeated at least three times with each experiment containing three replicates. The IC_{50} values were determined using Prism v.8 software.

4.2.4. Kinase selectivity assays

IC_{50} values of the synthesized compounds and TAE226 on five different kinases (FAK, FLT, FRK, IGF-1R, and BTK) were determined using Z′-LYTE® technology, which is based on FRET (Invitrogen/Life Technologies).

4.2.5. Western blotting

A-498 and Caki-1 cells were harvested, washed twice with chilled PBS, and lysed with ice-cold lysis buffer containing 0.1% SDS, 150 mM NaCl, 1 mM EDTA, 1% Triton X-100, 2 μg/ml aprotinin, 5 μg/ml pefabloc SC (4-(2-Aminoethyl) benzenesulfonyl fluoride hidrochloride), a protease inhibitor cocktail, 1% phosphatase inhibitor cocktail and 50 mM Tris—HCl (pH 7.4) after 24 h incubation with the IC_{50} value of the potent compounds (**5d**, **6a**, **11c**, and **7c**) or DMSO (control). The cell lysates were kept on ice for 30 min after gently vortex, and then centrifuged at 14,000g for 15 min at 4 °C. The supernatants were loaded onto SDS-PAGE gel immediately after protein extraction, or otherwise the supernatant was stored at −80 °C until use. The protein concentration was determined by Bradford assay (Bio-Rad Laboratories, Hercules, CA, USA) according to the manufacturer's instructions. For western blotting analysis, 40 μg protein-weight of cell lysate were loaded onto a SDS-PAGE gel. Proteins separated on a SDS-PAGE gel were transferred onto a PVDF membrane. The PVDF membrane was incubated in blocking buffer containing either 5% non-fat milk powder or 5% bovine serum albumin (Sigma-Aldrich) for 1 h. Subsequently, the PVDF membrane was incubated with primary antibodies for overnight and followed by HRP-conjugated anti-rabbit IgG (Cell Signaling Technologies) or HRP- conjugated anti-mouse IgG (Becton Dickinson Co, Durham, NC, USA) for 1 h with agitation at room temperature. Finally, the specific bindings to each primary antibody were detected on an X-ray film (Konica Minolta Medical Imaging, Wayne, NJ, USA) with ECL prime immunodetection reagent (GE Healthcare, Little Chalfont, UK).

4.2.6. Cell cycle assay

The flow cytometer was used to support the anti-proliferative activity of the synthesized compounds towards A-498 and Caki-1 cells. Briefly, cells at a density of 1×10^5 cells/mL were cultured with different concentrations of **6a** (0.75 and 1.5 μM) or DMSO as a negative control and incubated for 24 h. The cells were harvested and washed with PBS. Subsequently, 80% ice cold ethanol was added to the cell pellets drop by drop with continuous vortexing to prevent clumping and aggregation of cells and then kept at 20∘C for overnight. Then, the cell pellets washed twice, stained with PI, and incubated in the dark for 20 min. Flow cytometric analysis was conducted using FACS Calibur flow cytometer (BD Biosciences, San Jose, CA, USA).

4.2.7. Apoptosis assay

Induction of apoptosis of A-498 and Caki-1 cells was analyzed using the annexin V/propidium iodide (PI) staining kit (BioLegend, San Diego, CA, USA) according to the manufacturer's instructions. A-498 and Caki-1 cells were treated with different concentrations

of **6a** (0.75 and 1.5 μM) or DMSO as a negative control and incubated for 24 h. Flow cytometric analysis was conducted using FACS Calibur flow cytometer (BD Biosciences, San Jose, CA, USA). Quantitative analysis of the FACS data was performed with FlowJo software (FlowJo, Ashland, OR, USA).

4.2.8. Resistance development assays

To examine whether A-498 cells can develop resistance to 6a or valproic acid, A-498 cells were exposed to 30% of IC$_{50}$ of **6a** or valproic acid for 12 weeks, followed by treatment with different concentrations of 6a (0.25, 0.5, 1, 5, and 10 μM) or valproic acid (VPA; 0.25, 0.5, 1, 2, and 4 mM) or DMSO as a negative control for 24 h. Then, the cells were subjected to the MTT assay to detect cell growth inhibition or to the Western blot assay to detect HDAC2, p-FAK, FAK, p-Akt, and Akt proteins expression. Control cell cultures were not pretreated with 6a or VPA.

4.2.9. Determination of caspase-8 & caspase-9

The determination of caspase-8 and caspase-9 was performed using human caspase-8 ELISA kit (EIA-4863) and human caspase-9 ELISA kit DRG® (EIA-4860) (DRG International Inc., USA). Cells were lysed with cell extraction buffer. The lysate was diluted in standard diluent buffer over the range of the assay and measured for human caspase-8 and caspase-9 content, separately. For each assay, the cells were plated in a density of 1.2−1.8 × 10,000 cells/well in a volume of 100 μl complete growth medium +100 μL of the tested compound per well in a 96-well plate for 24 h before the enzyme assay for caspase-8 and caspase-9 according to the manufacturer's protocol. Finally, the absorbance of each microwell was read using a microplate reader set at 450 nm.

4.2.10. Determination of caspase-3

The level of human active caspase-3 protein was evaluated using Invitrogen (Catalog KHO1091) ELISA kit. The manufacturer's instructions were followed in the following procedures. Add 100 μL of the standard diluent buffer to the zero standard wells. Add 100 μL of standards and controls or diluted samples to the appropriate microtitre wells. Cover wells and incubate for 2 h at room temperature. Thoroughly aspirate or decant solution from wells and discard the liquid. Pipette 100 μL of caspase-3 (active) detection antibody solution into each well. Cover plate and incubate for 1 h at room temperature. Add 100 μL Anti-Rabbit IgG HRP working solution to each well. Cover wells with the plate cover and incubate for 30 min at room temperature. Add 100 μL of stabilised chromogen to each well. The liquid in the wells will begin to turn blue. Incubate for 30 min at room temperature. Stop solution has been added to each well. The solution in the wells should change from blue to yellow. Read the plate within 2 h after adding the stop solution. Use a curve fitting software to generate the standard curve.

4.2.11. Docking study

Molecular modeling and visualization processes were performed within HDAC2 and FAK active sites using Molecular Operating Environment (MOE) 2019.0102 software (Chemical Computing Group, Montreal, QC, Canada). The co-crystal structures were retrieved from the RCSB Protein Data Bank (PDB codes 4LXZ and 2JKK). First, **6a** was prepared with the standard protocol designated in MOE 2019. However, the **6a** structure's energy was minimized using Amber10:EHT forcefield with a gradient RMSD of 0.0001 kcal/mol. Then the protein structures were prepared by using the MOE QickPrep protocol. To validate the docking study at the binding sites, the native ligands were re-docked into the binding site using the same set of parameters as described above. The RMSD values of the best-docked poses were 0.8172 Å (SAHA) and 0.8762 Å (TAE226) in HDAC2 and FAK active sites, respectively,

thus validating the docking using MOE. Compound **6a** was then docked into the binding sites using the Alpha triangle placement method. Refinement was carried out using Forcefield and scored using the Affinity dG scoring system. The resulting docking poses were visually inspected, and the pose of the lowest binding free energy value was considered.

5. Statistical analysis

Data are represented as means ± SEM. Statistical significance was assessed using one-way analysis of variance (ANOVA) including a Tukey's test for multiple comparisons. *p < 0.05, **p < 0.01, and ***p < 0.001. The data shown in the figures are representative data for three independent experimental results.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.ejmech.2021.113569.

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